

# HOW SIDE REACTIONS CAN INFLUENCE POLY(2-OXAZOLINE) SYNTHESIS FOR POLYMER THERAPEUTICS AND HYDROGELS

F. J. Arraez,<sup>1</sup> X. Xu,<sup>2</sup> P.H.M. Van Steenberge,<sup>1</sup> R. Hoogenboom,<sup>1</sup> D. R. D'hooge<sup>1,3</sup>

<sup>1</sup>Laboratory for Chemical Technology (LCT), Ghent University

<sup>2</sup>Supramolecular Chemistry (SC) group, Ghent University

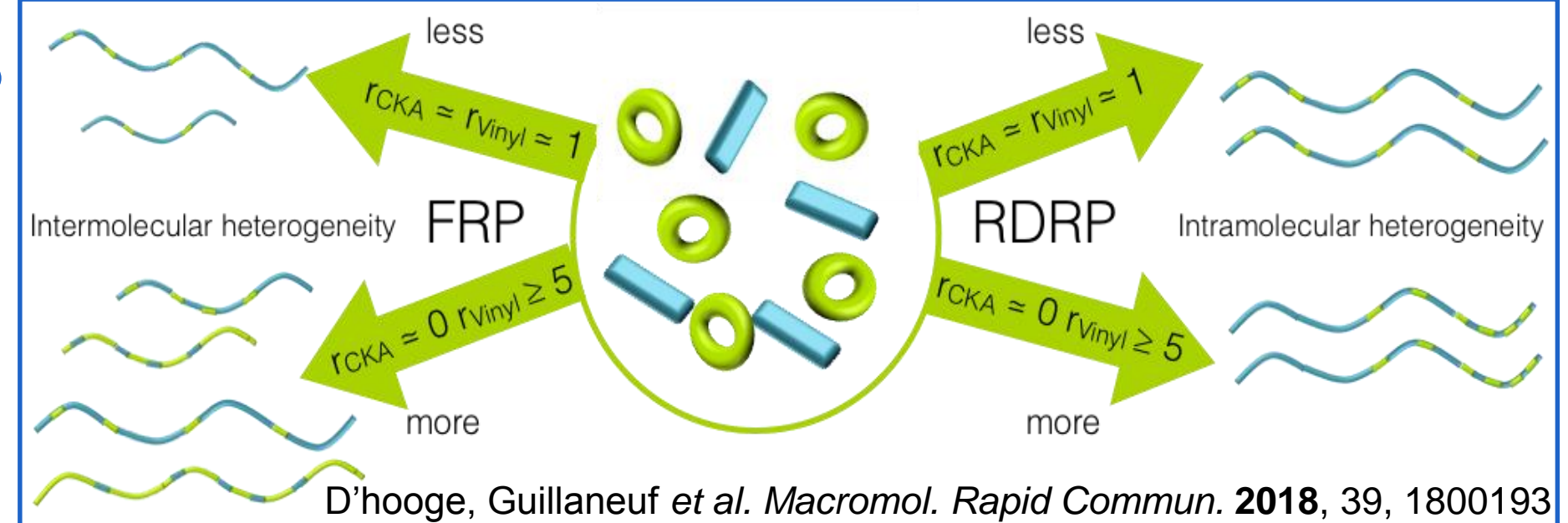
<sup>3</sup>Centre for Textile Science and Engineering (CTSE), Ghent University



The diagram illustrates the self-assembly of a fructose-coated mixed micelle and its passive targeting to tumor tissue. At the top, the components are shown: PBA (green dots), Drug (red dots), Fructose (yellow dots), and Glycoprotein (blue dots). These components undergo self-assembly to form a micelle. The micelle is then shown interacting with a normal cell, where it is rejected due to 'Inhibit adsorption' and 'Avoid recognition'. The micelle then enters a tumor tissue, where it is taken up by a cell via 'Receptor mediated cellular uptake'. The tumor tissue is characterized by a pH of 6.0-7.0, which causes the PBA to be exposed, leading to the micelle's disassembly and the release of the drug into the nucleus.

**Legend:**

- PBA
- Drug
- Fructose
- Cis-diol-containing components
- SA
- Glycoprotein



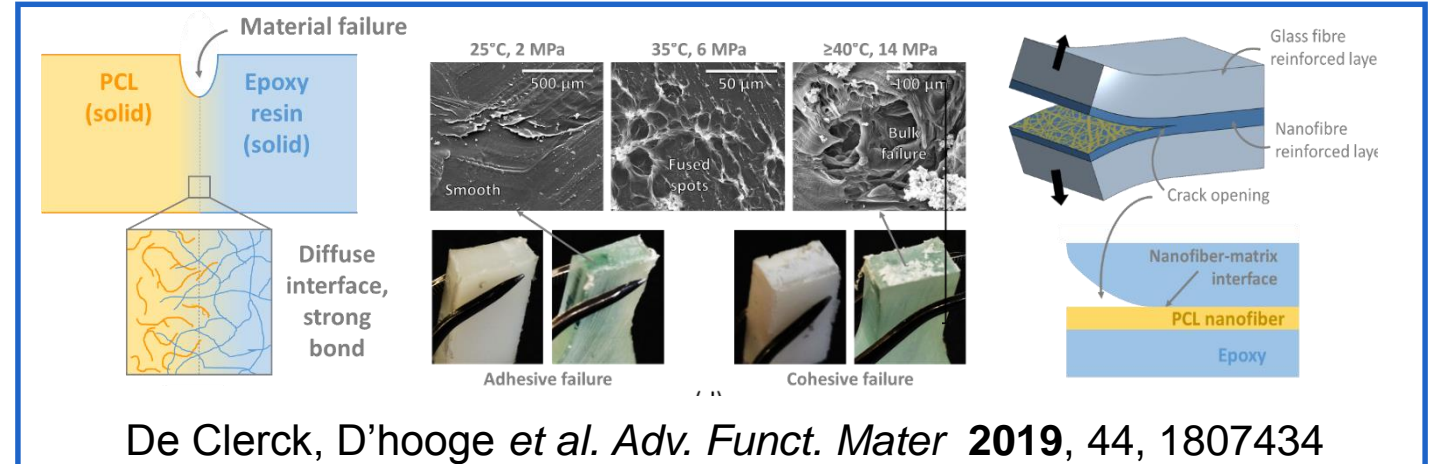
The diagram illustrates the degradation of the EVA encapsulation in a solar cell. The structure consists of a glass front layer, a Si cell with an Ag contact, an EVA encapsulation layer, and a backsheet. UV light is shown incident on the glass. The diagram illustrates the formation of radicals ( $k_1, k_2, k_3, k_4$ ) and crosslinking in the EVA, leading to hydrolytic depolymerization and the release of  $O_2$  and  $H_2O$ . Insets show the chemical structures of Ac (acetyl) groups and bulk EVA (polyethylene glycol).



*selected monomers   programmed polymerization   structure and function*

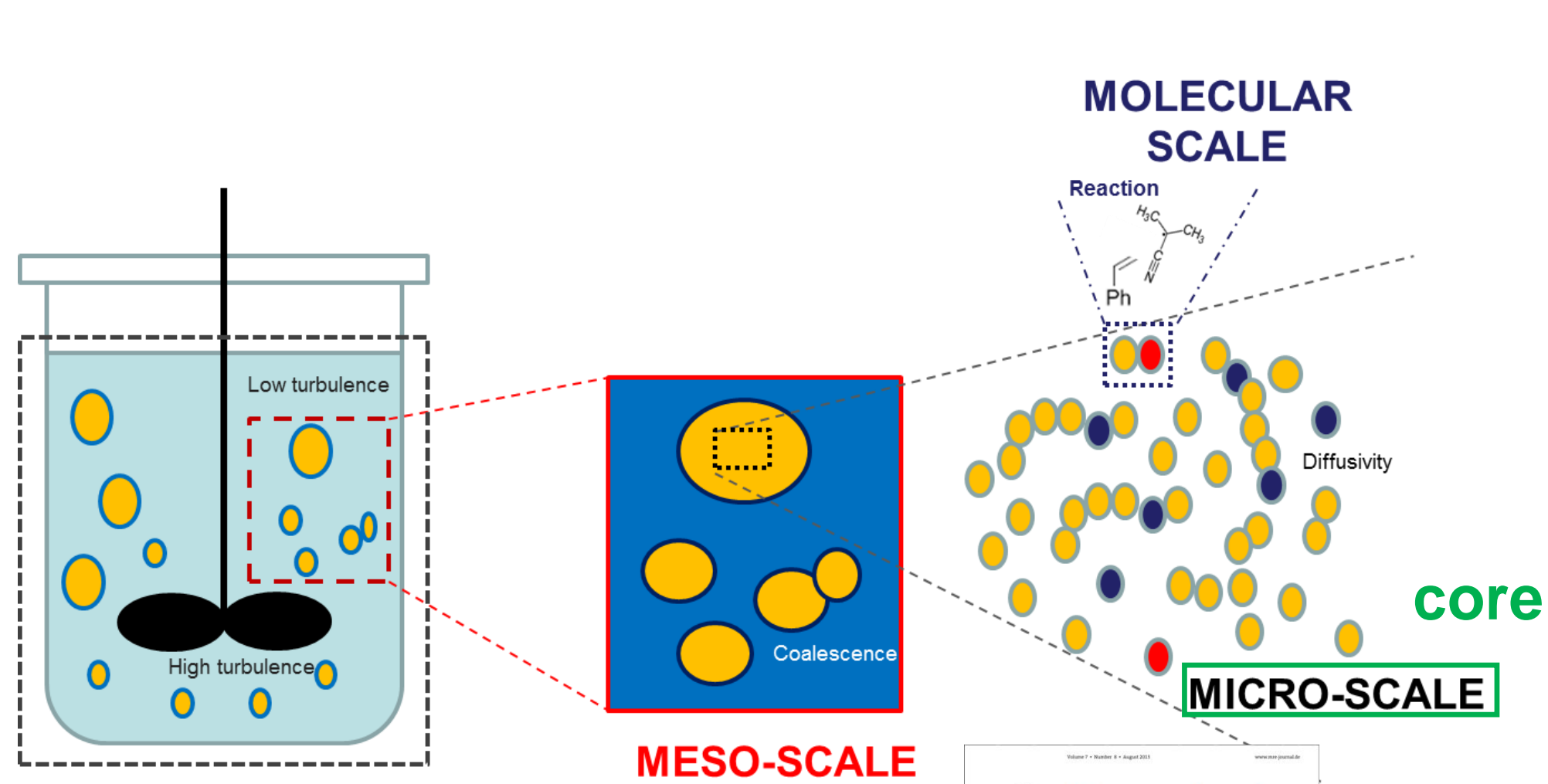
Szymanski *et al.* *Acc. Chem. Res.* **2018**, 51, 649

D'hooge, Lutz *et al.* *Macromolecules* **2016**, 49, 9336





# MULTI-SCALE CHARACTER OF POLYMERIZATION

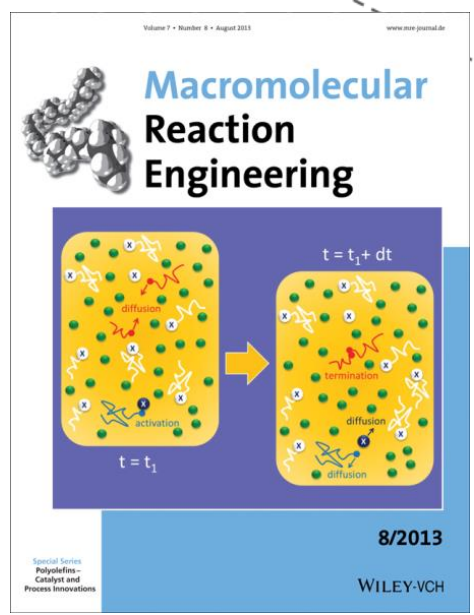


MACRO-SCALE

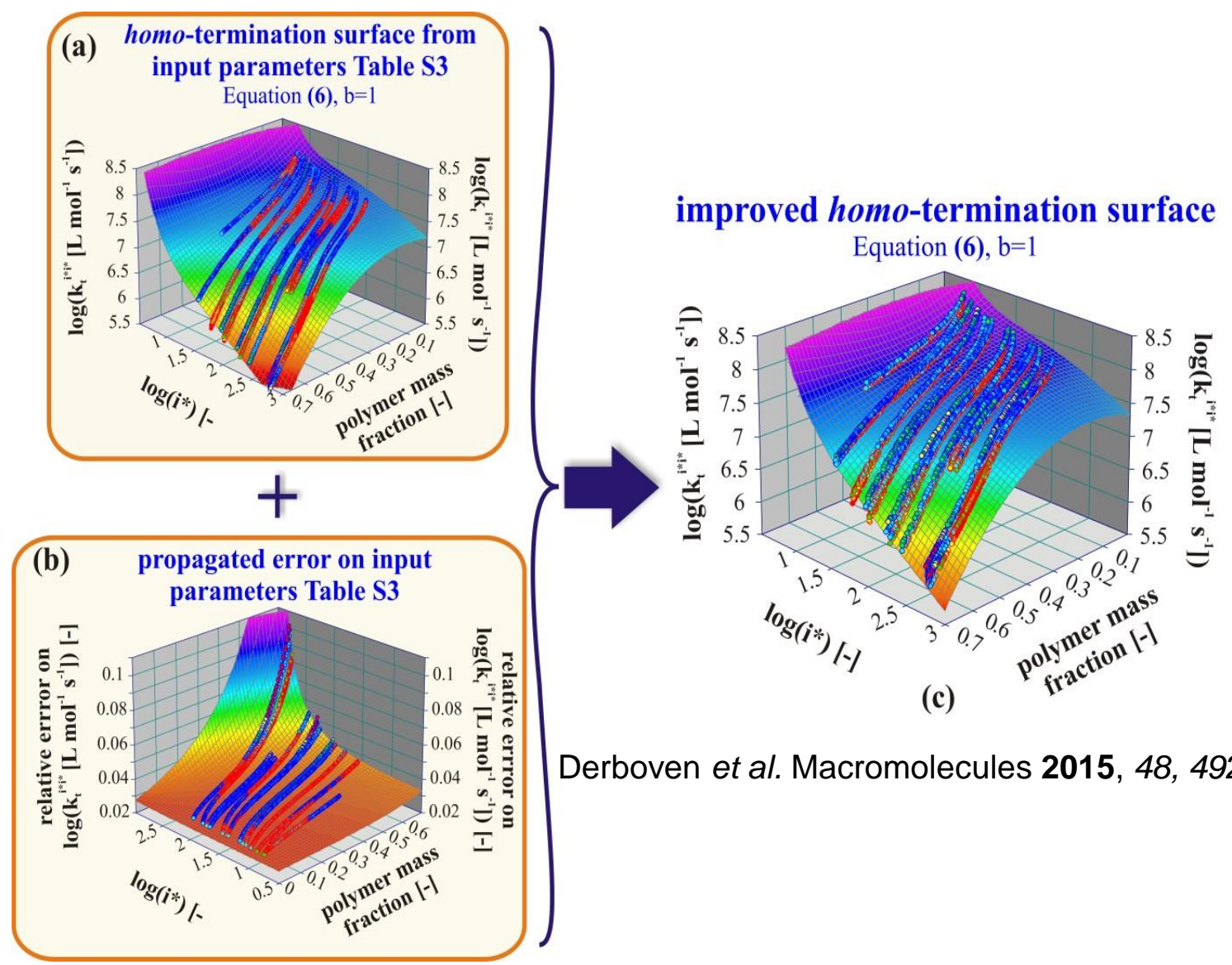
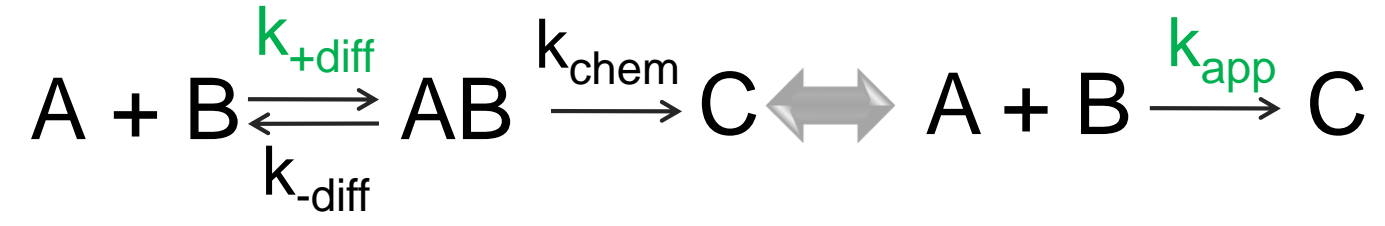
MESO-SCALE

MICRO-SCALE

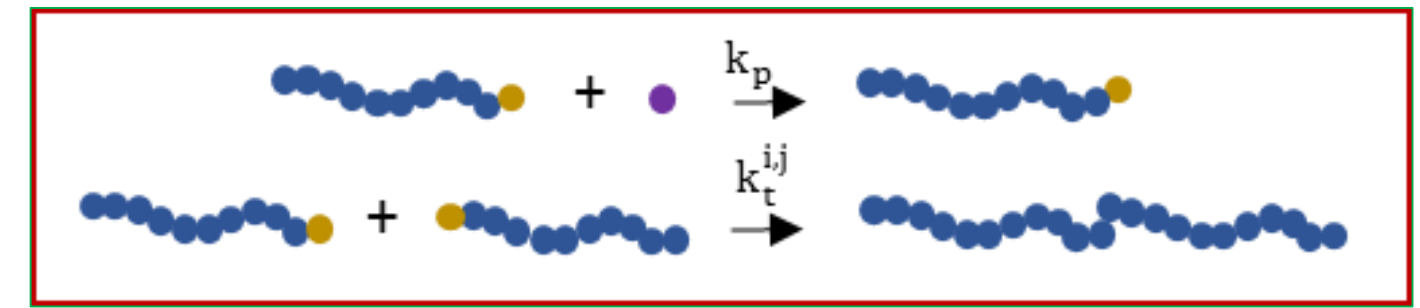
D'hooge D.R. *et al. Prog. Polym. Sci.* **2016** 58, 59  
Mastan *et al. Prog. Polym. Sci.* **2015**, 45, 71



D'hooge *et al. Macromol. React Eng.* **2013**, 7, 362  
Peklak *et al. J. Polym. Sc. Polym Chem/* **2006**, 44, 1071  
Achilias and Kiparissides *Macromolecules* **1992**, 25, 3739

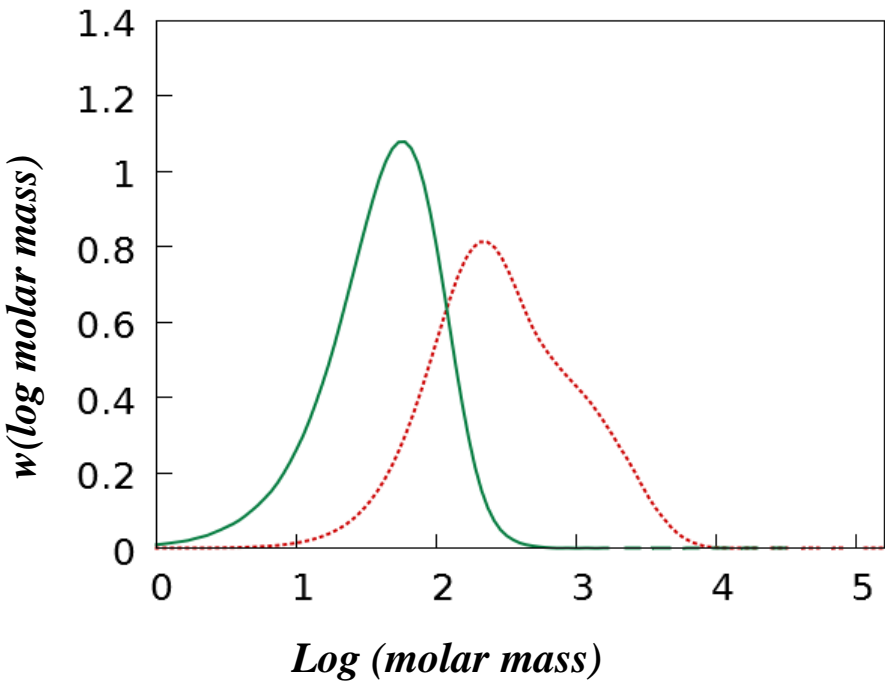


Derboven *et al. Macromolecules* **2015**, 48, 492

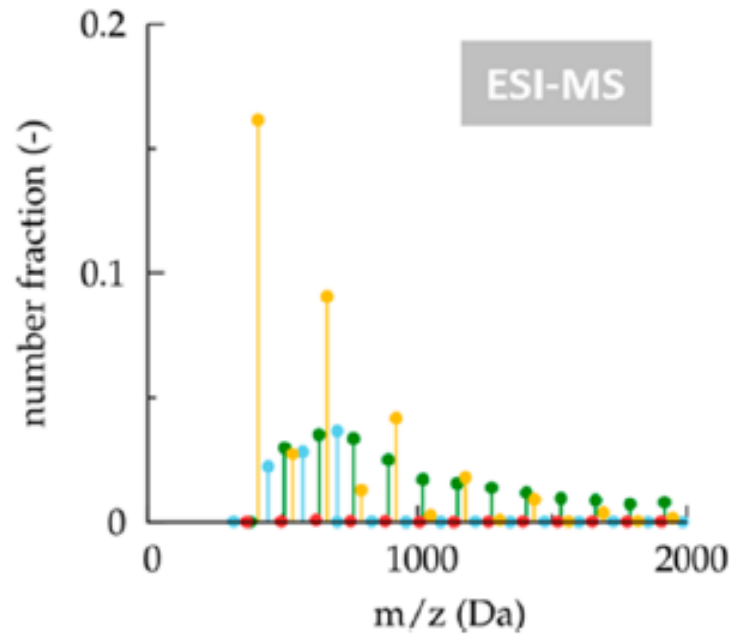


# MICRO-SCALE: SYNERGY MODEL AND EXPERIMENT

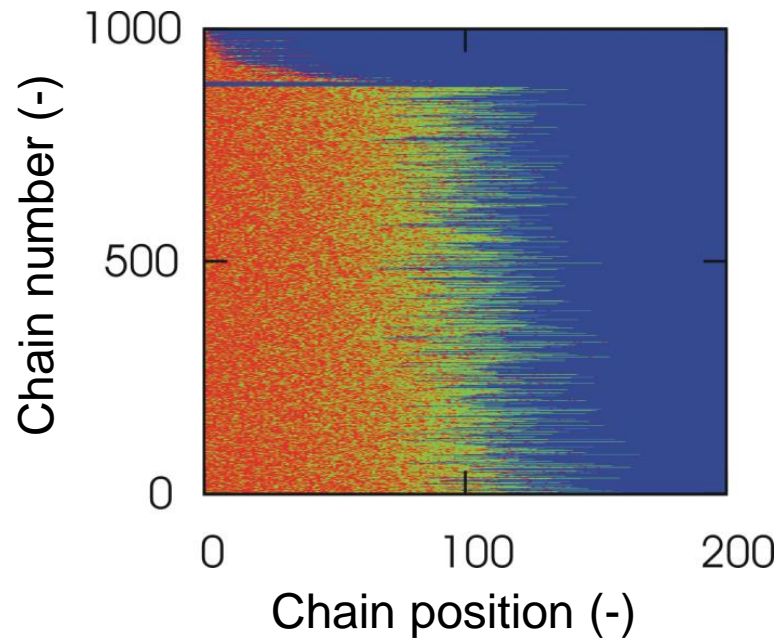
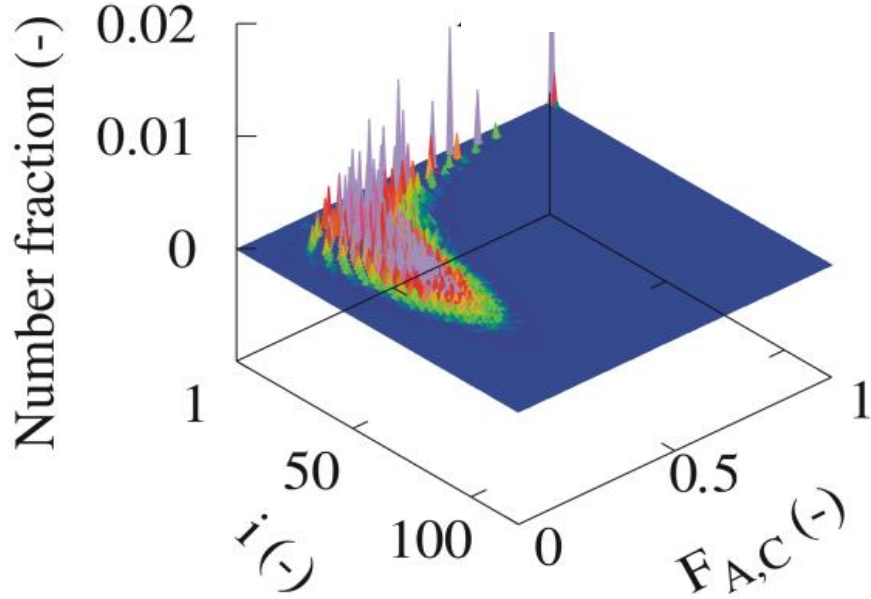
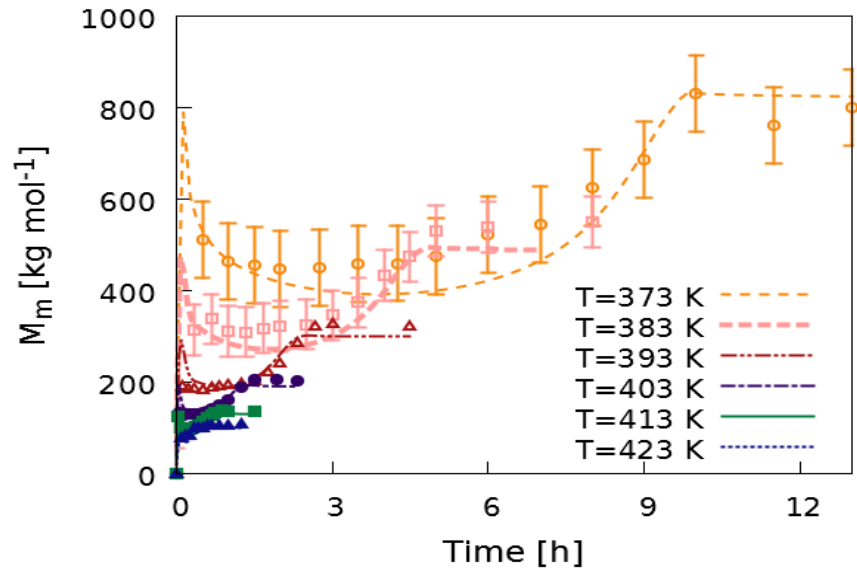
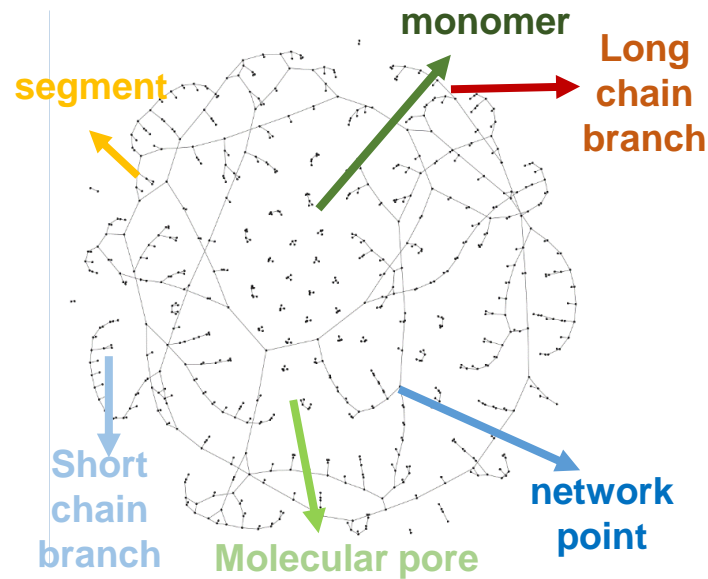
Generation 1



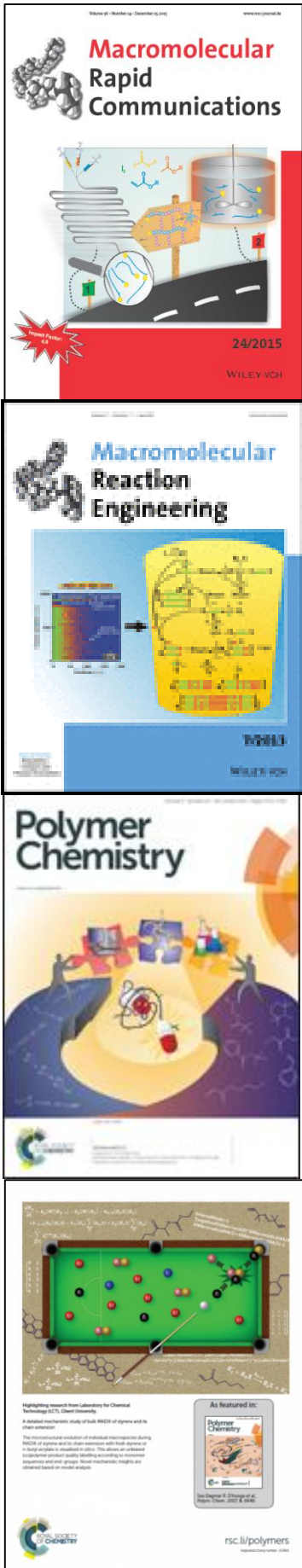
Generation 2



Generation 3



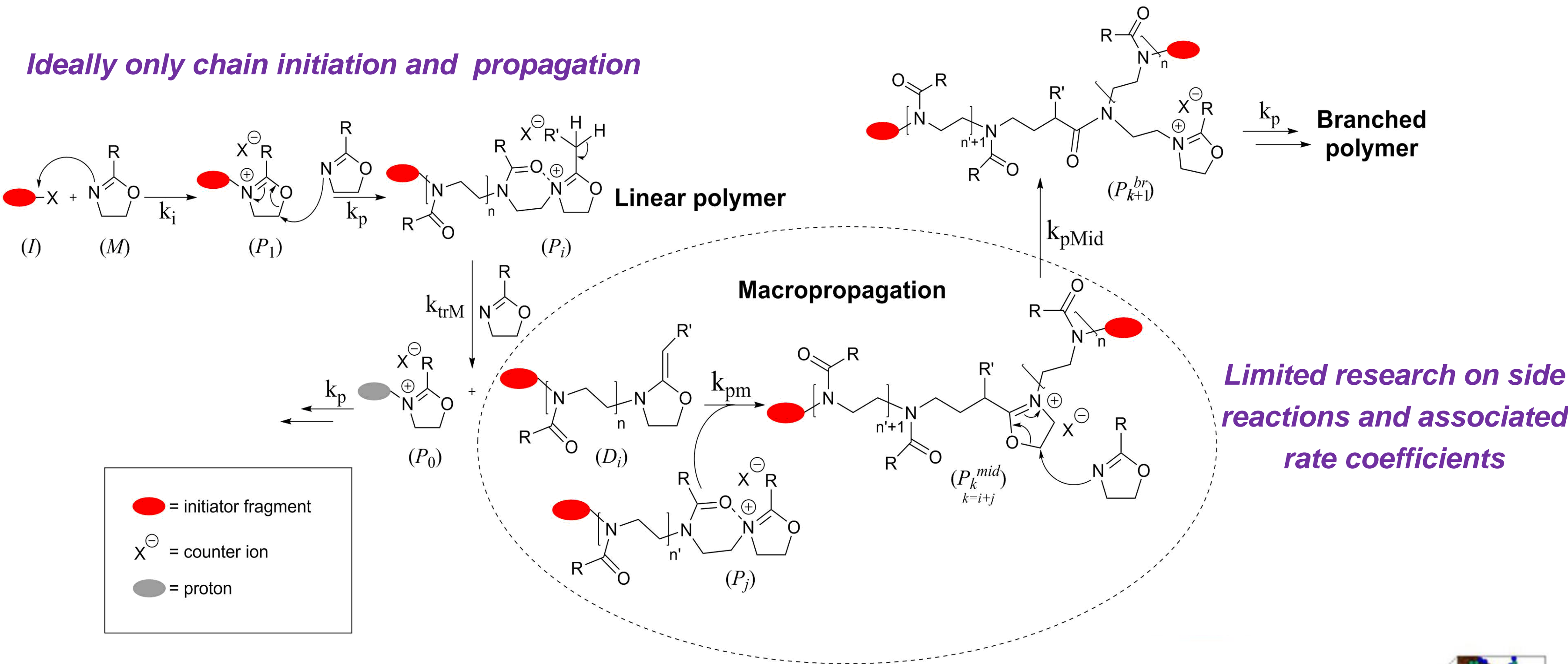
*Complexity and macromolecular structural detail*





# CATIONIC RING-OPENING POLYMERIZATION OF 2-OXAZOLINES

*Ideally only chain initiation and propagation*

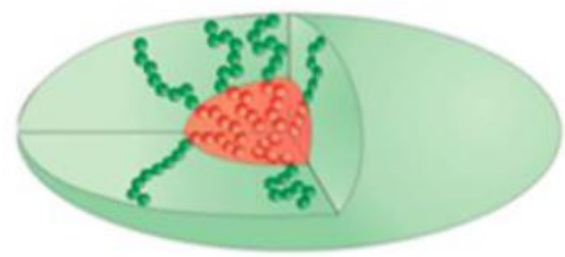


*Solution polymerization*

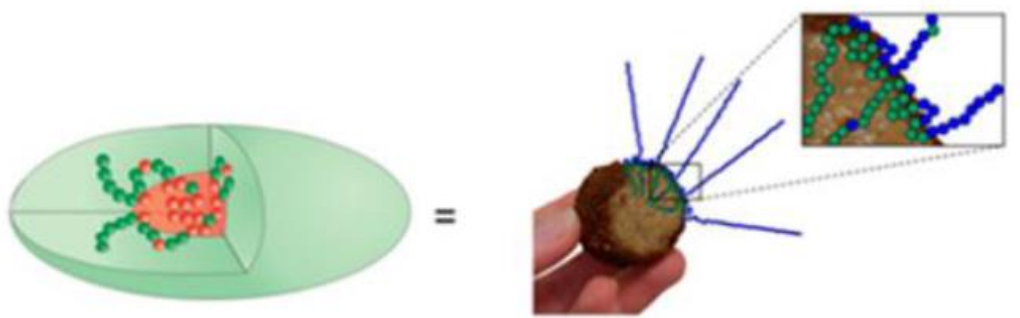


*Main focus on chemistry*

**block:**



**gradient:**



# OUTLINE

## 1. Homopolymerization

- Chain initiation and propagation reactivity
- Chain transfer to monomer reactivity
- Macropropagation reactivity

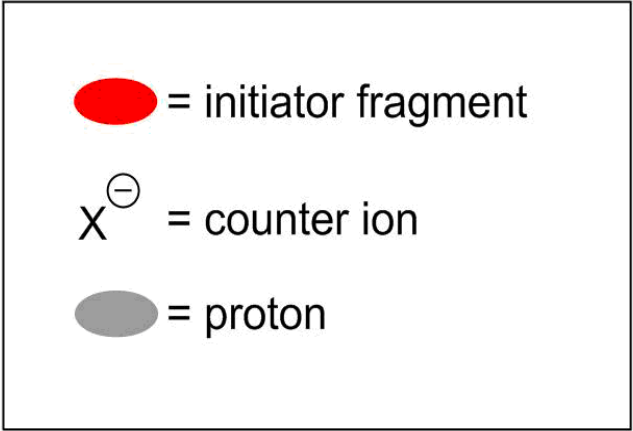
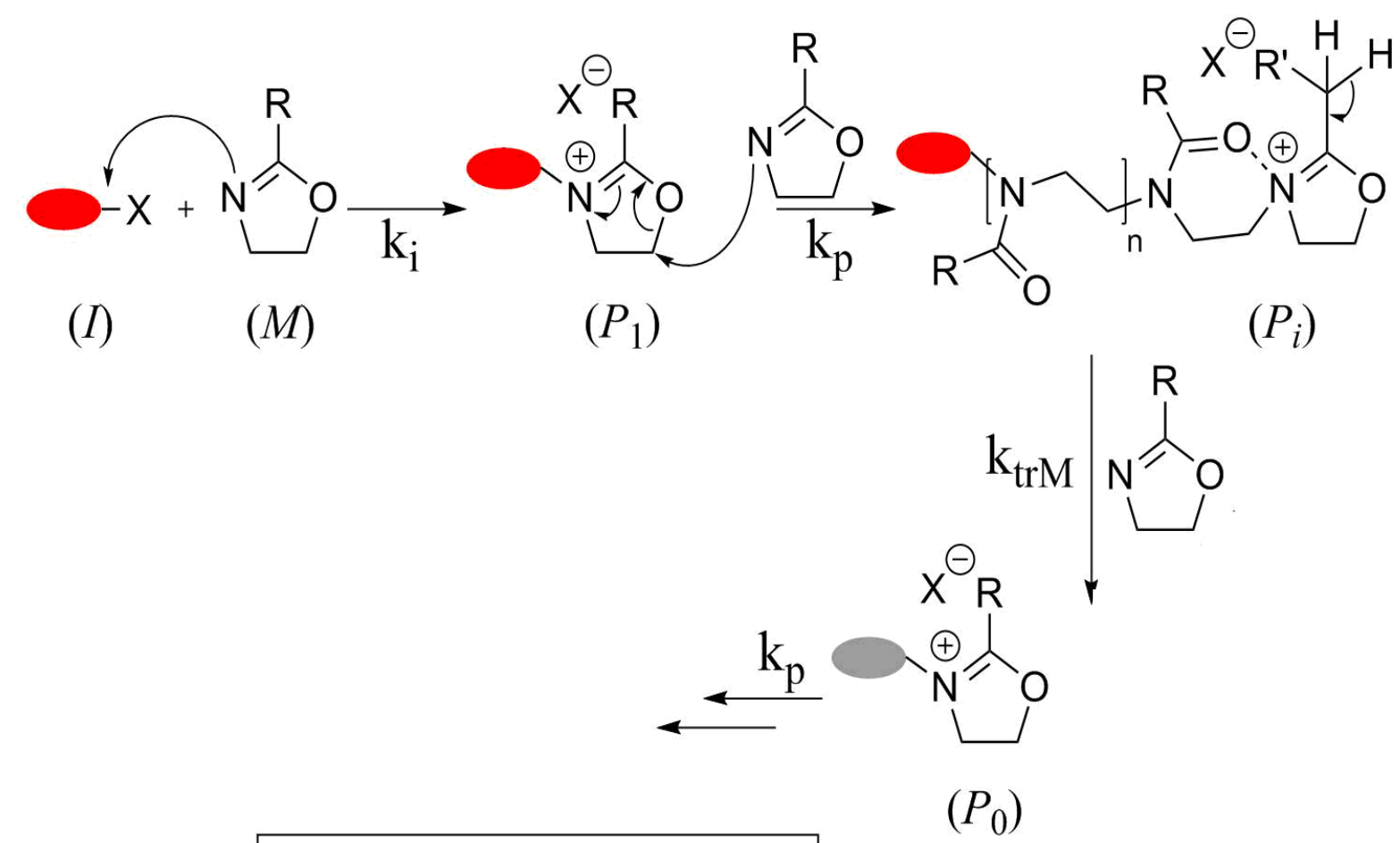
Van Steenberge *et al. Macromolecules* **2015**, 48, 7765  
Glassner *et al. Eur. Polym. J.*, **2015**, 65, 298  
Van Steenberge *et al. Nat. Commun.* **2019**, to be subm.  
Arraez *et al. Macromolecules* **2019**, to be submitted

## 2. Copolymerization with functional comonomer in view of therapeutics & hydrogels

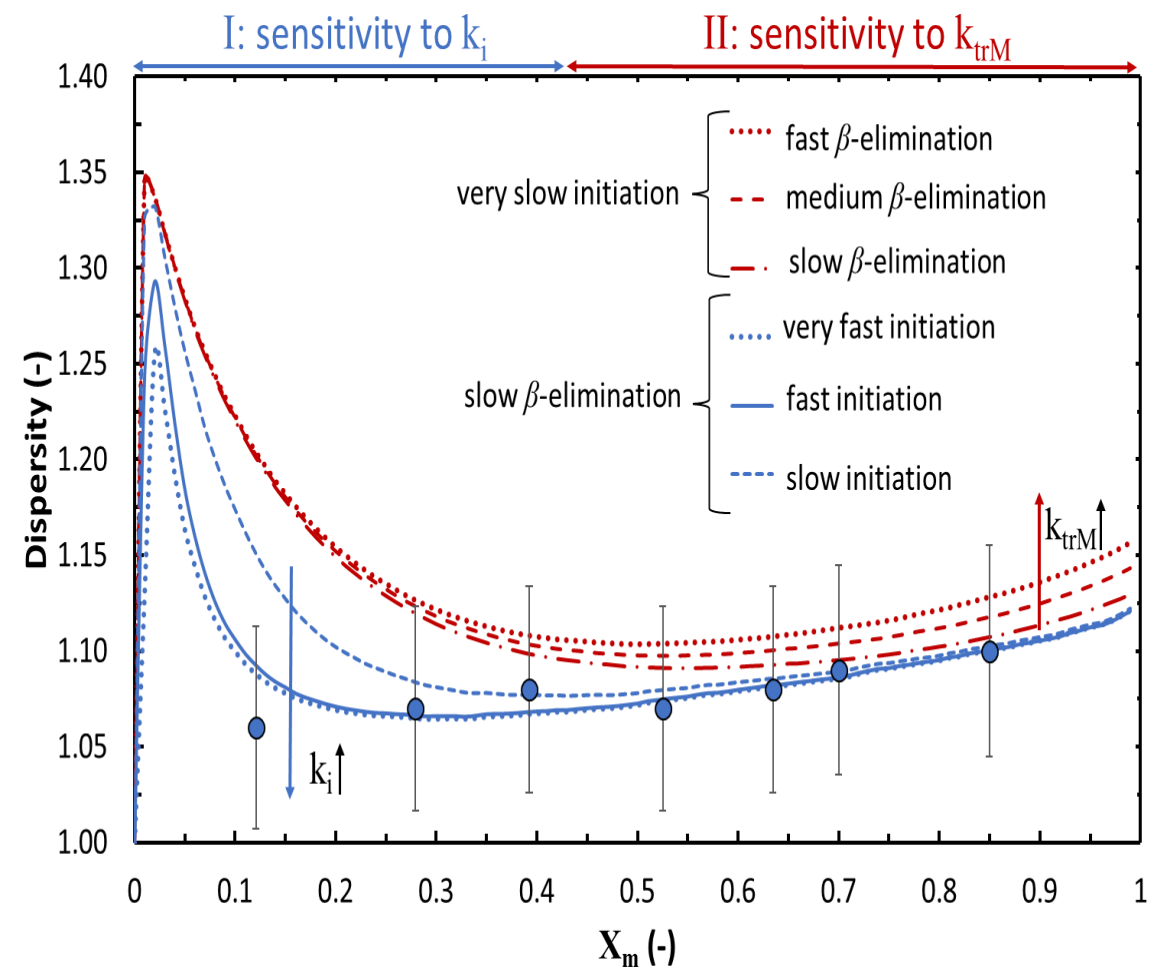
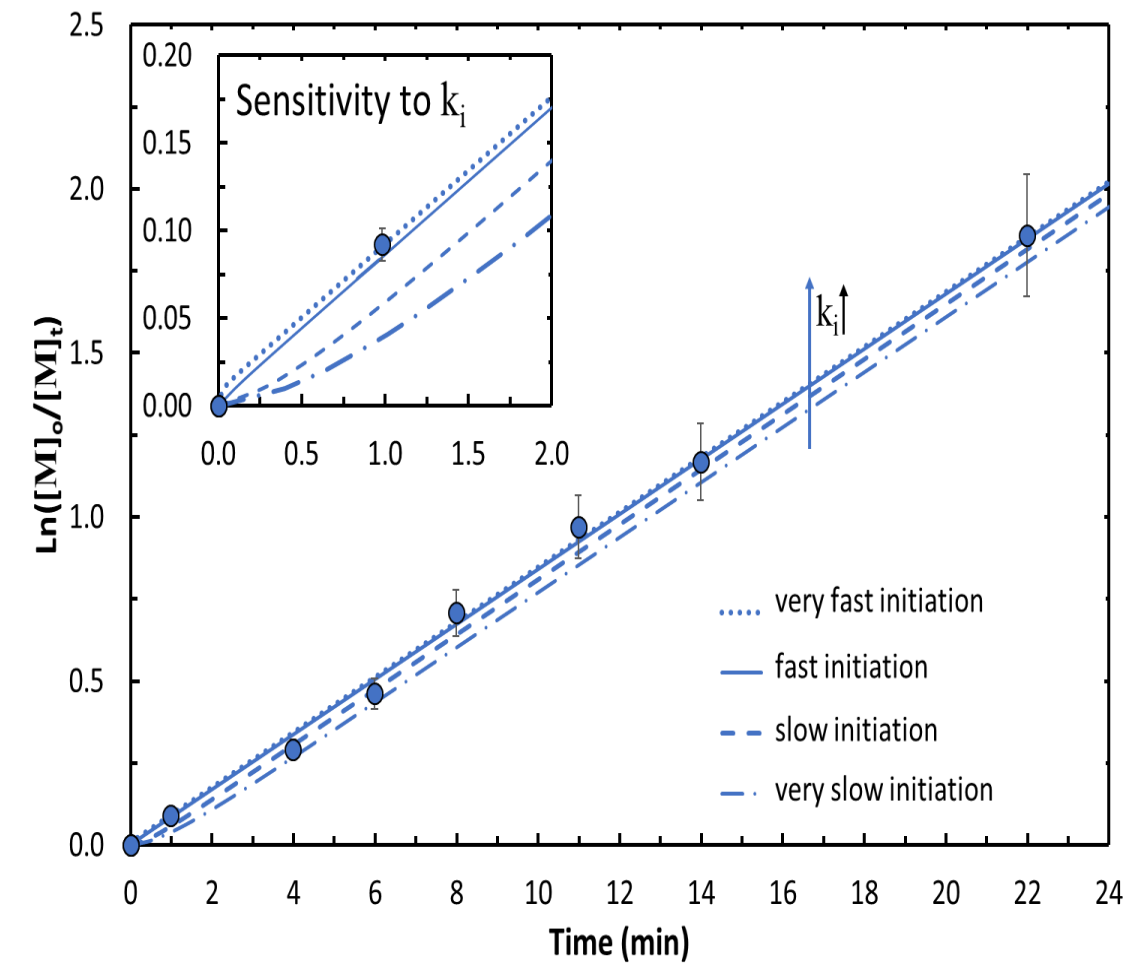
- Unbiased qualification tool for evaluation functionalization success
- Design of chemical structure of the comonomers and reaction conditions
- Applicability for low and high target degrees of polymerization (target DPs)

Van Steenberge *et al. Macromolecules* **2015**, 48, 7765  
Van Steenberge *et al. Nat. Commun.* **2019**, to be subm.

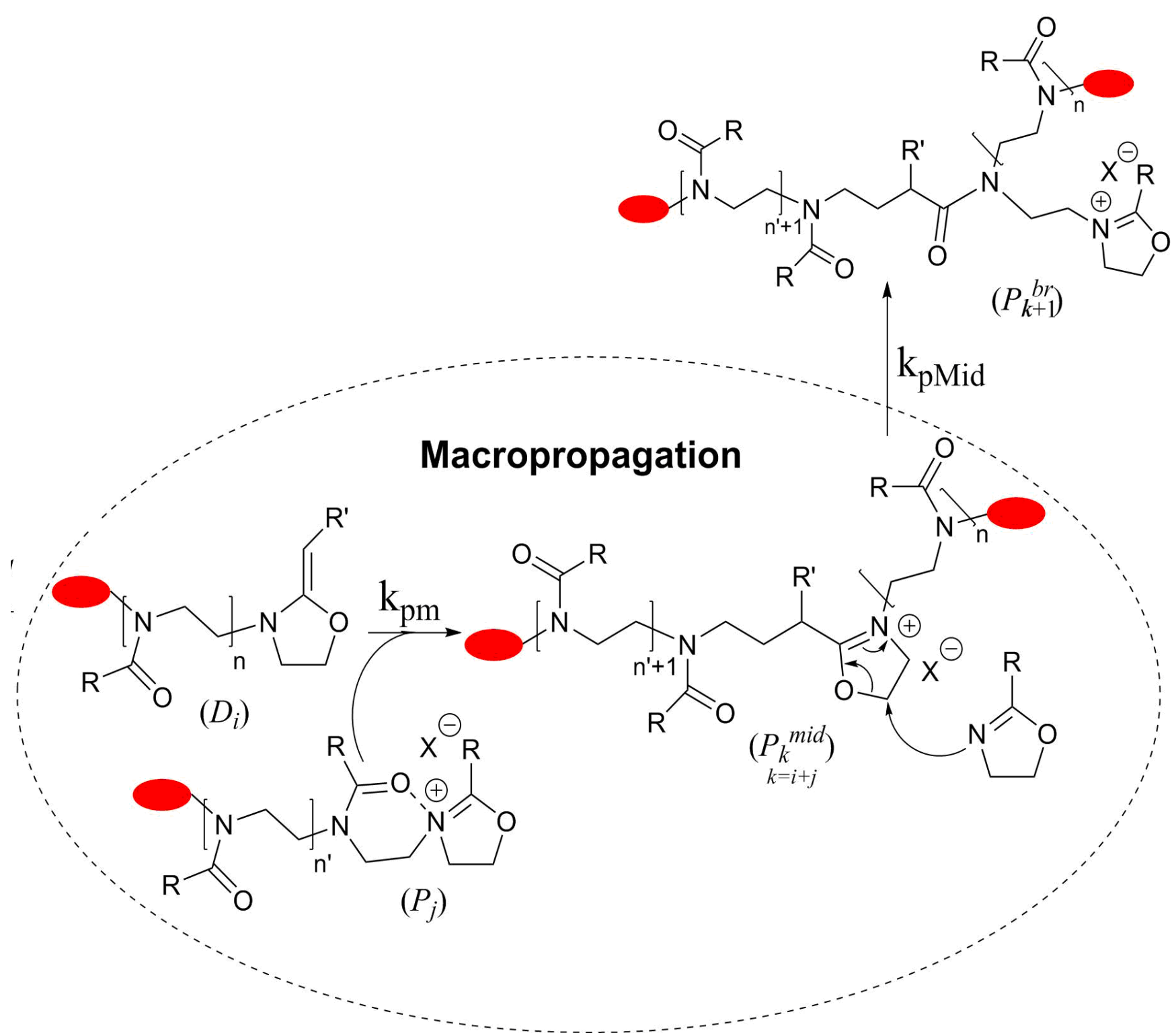
# WHICH EXPERIMENTAL RESPONSES ?



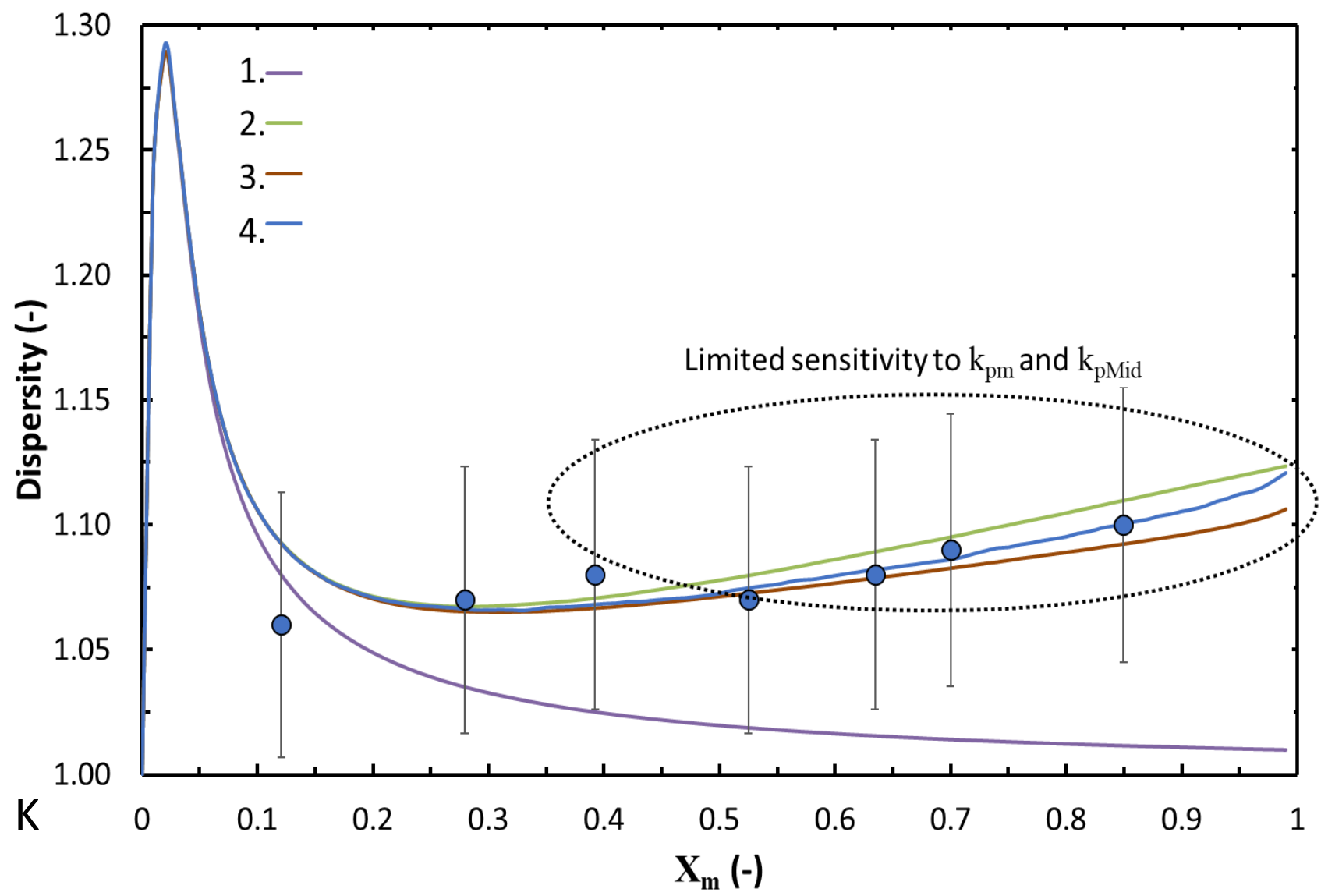
CROP of EtOx ( $[M]_0 = 4 \text{ mol}\cdot\text{L}^{-1}$ ; target DP of 100; acetonitrile; 393 K



# WHICH EXPERIMENTAL RESPONSES ?



	$k_{trM}$ (L·mol <sup>-1</sup> ·s <sup>-1</sup> )	$k_{pm}$ (L·mol <sup>-1</sup> ·s <sup>-1</sup> )	$k_{pMid}$ (L·mol <sup>-1</sup> ·s <sup>-1</sup> )
1. —	✗	✗	✗
2. —	✓ 6.57×10 <sup>-2</sup>	✗	✗
3. —	✓ 6.57×10 <sup>-2</sup>	✓ 6.99×10 <sup>-2</sup>	✗
4. —	✓ 6.57×10 <sup>-2</sup>	✓ 6.99×10 <sup>-2</sup> ×0.1	✓ 3.49×10 <sup>-2</sup>
5. —	✓ 6.57×10 <sup>-2</sup>	✓ 6.99×10 <sup>-3</sup>	✓ 3.49×10 <sup>-2</sup> ×10
6. - - -	✓ 6.57×10 <sup>-2</sup>	✓ 6.99×10 <sup>-3</sup>	✓ 3.49×10 <sup>-1</sup> ×10
●			
Experimental			

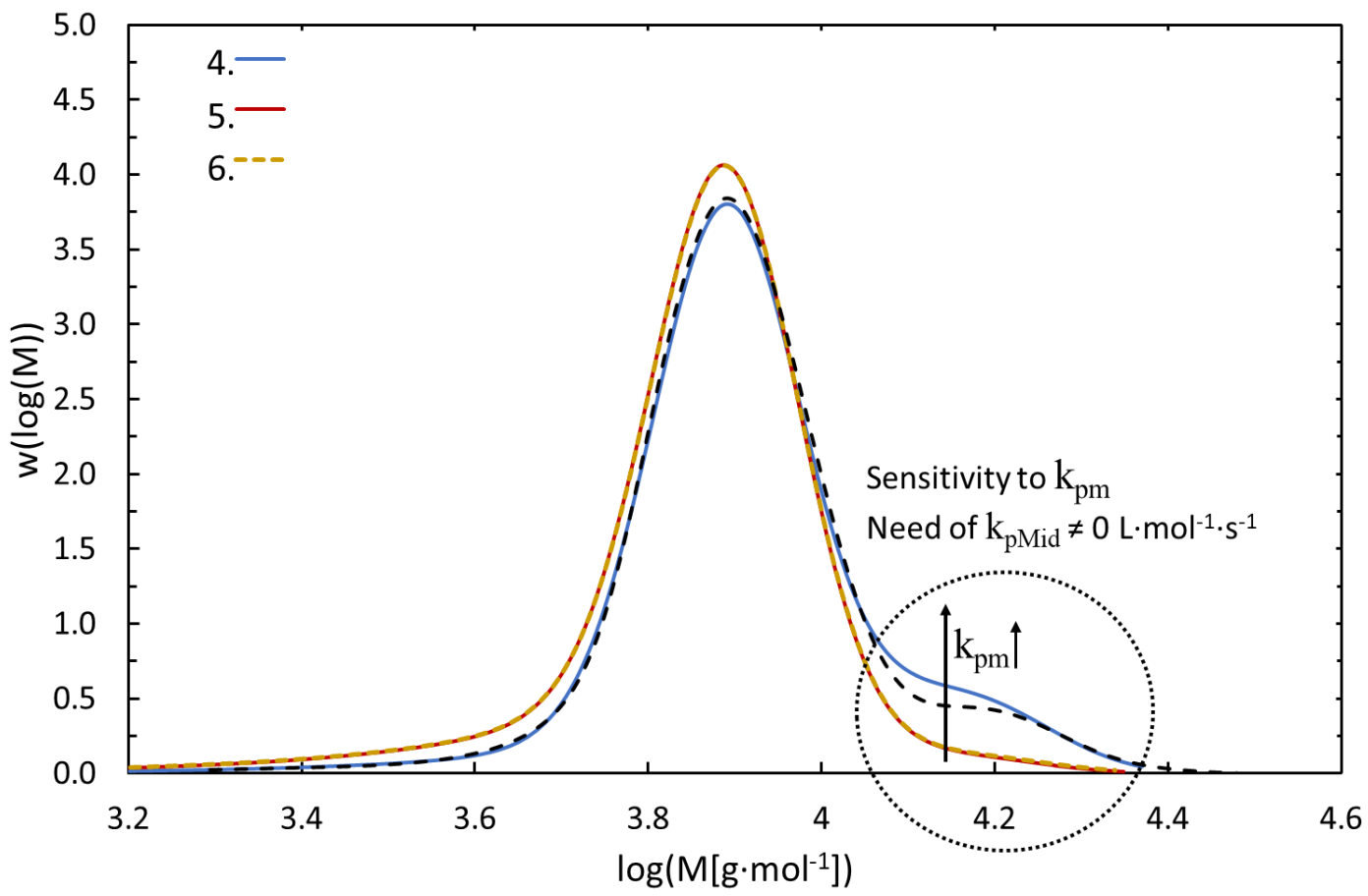
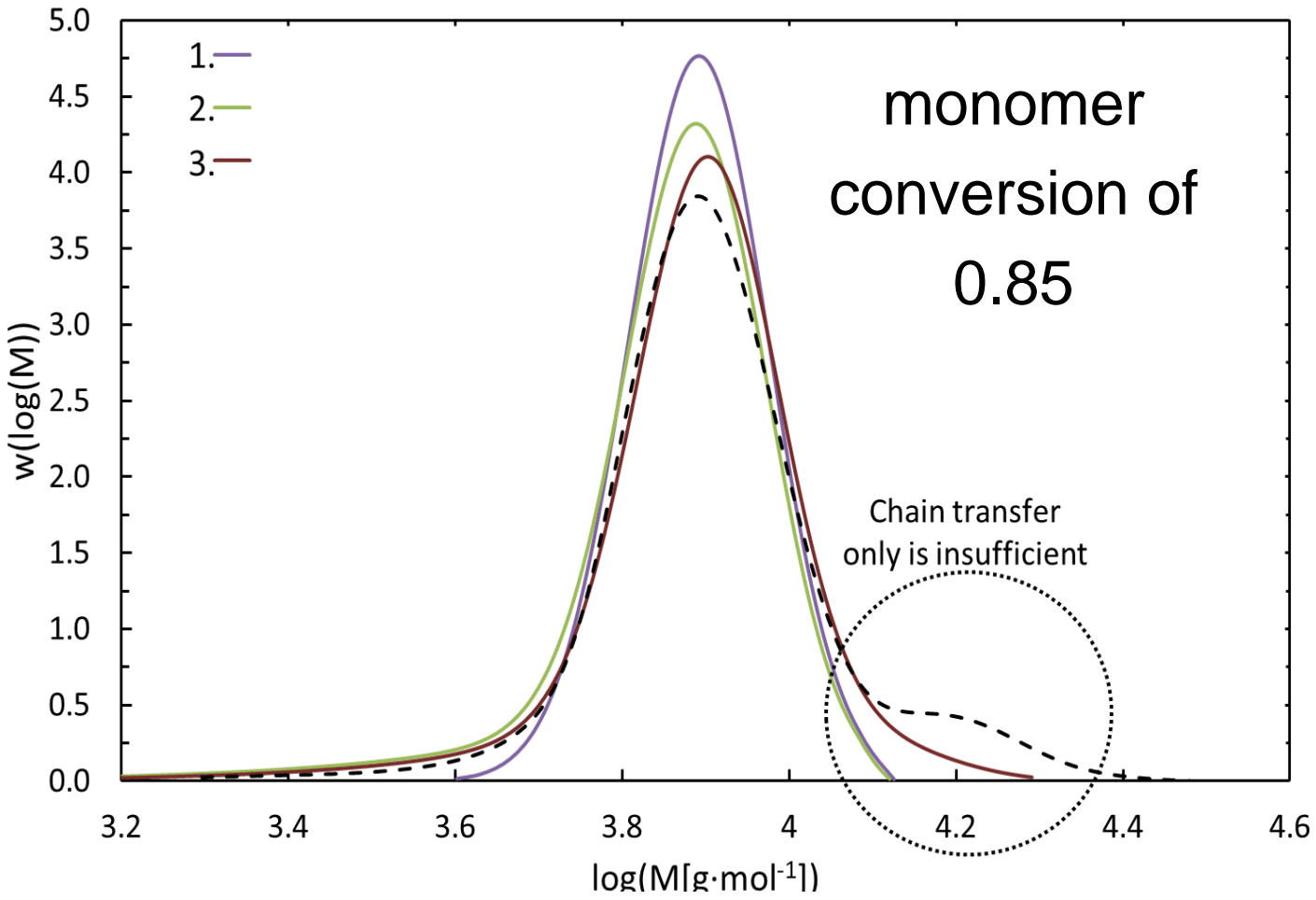
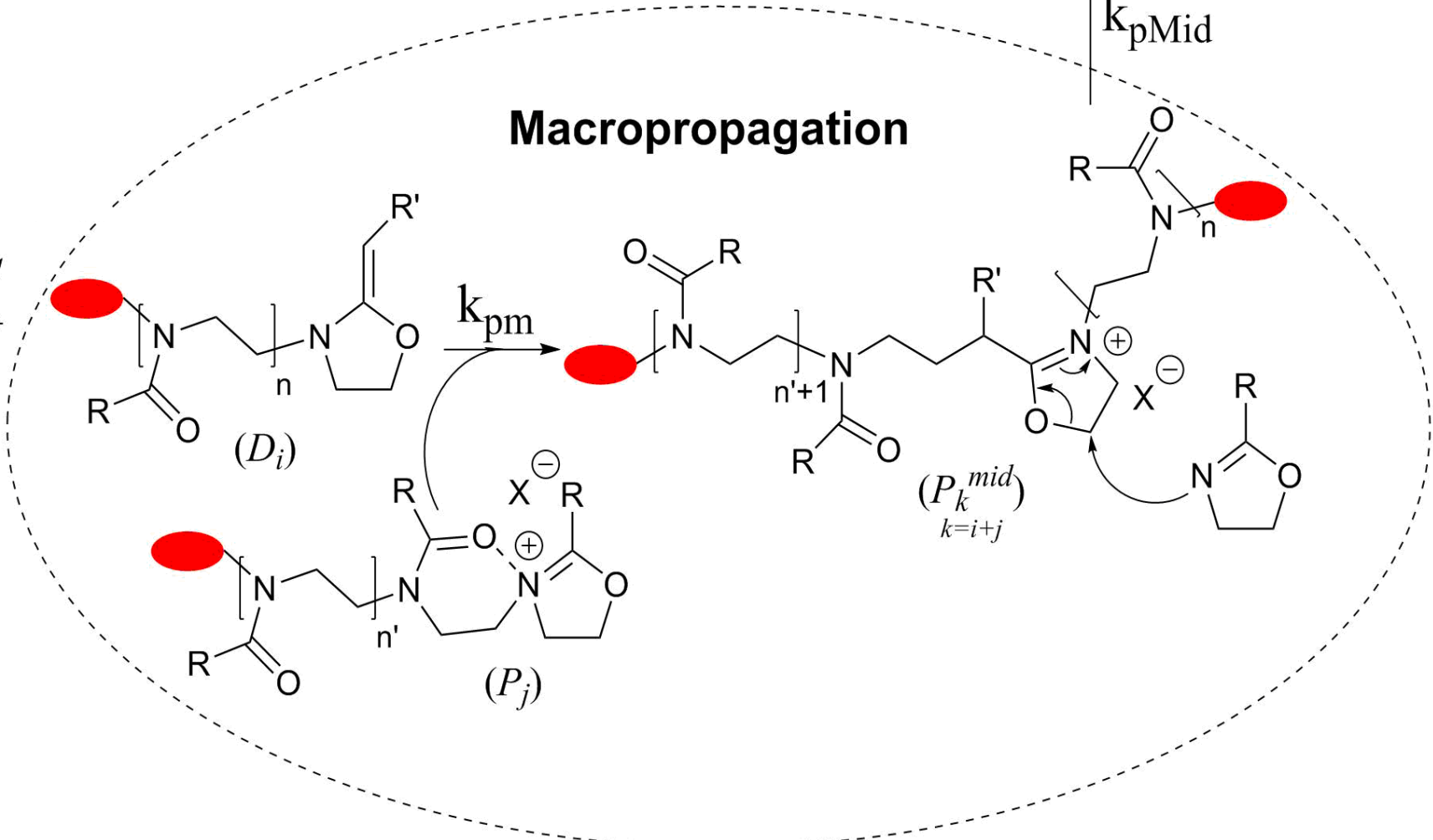
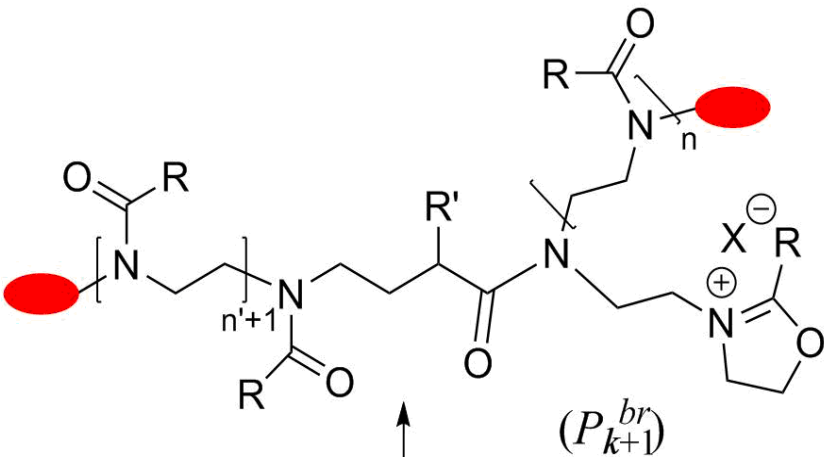


CROP of EtOx ( $[M]_0 = 4 \text{ mol}\cdot\text{L}^{-1}$ ; target DP of 100; acetonitrile; 393 K



# WHICH EXPERIMENTAL RESPONSES ?

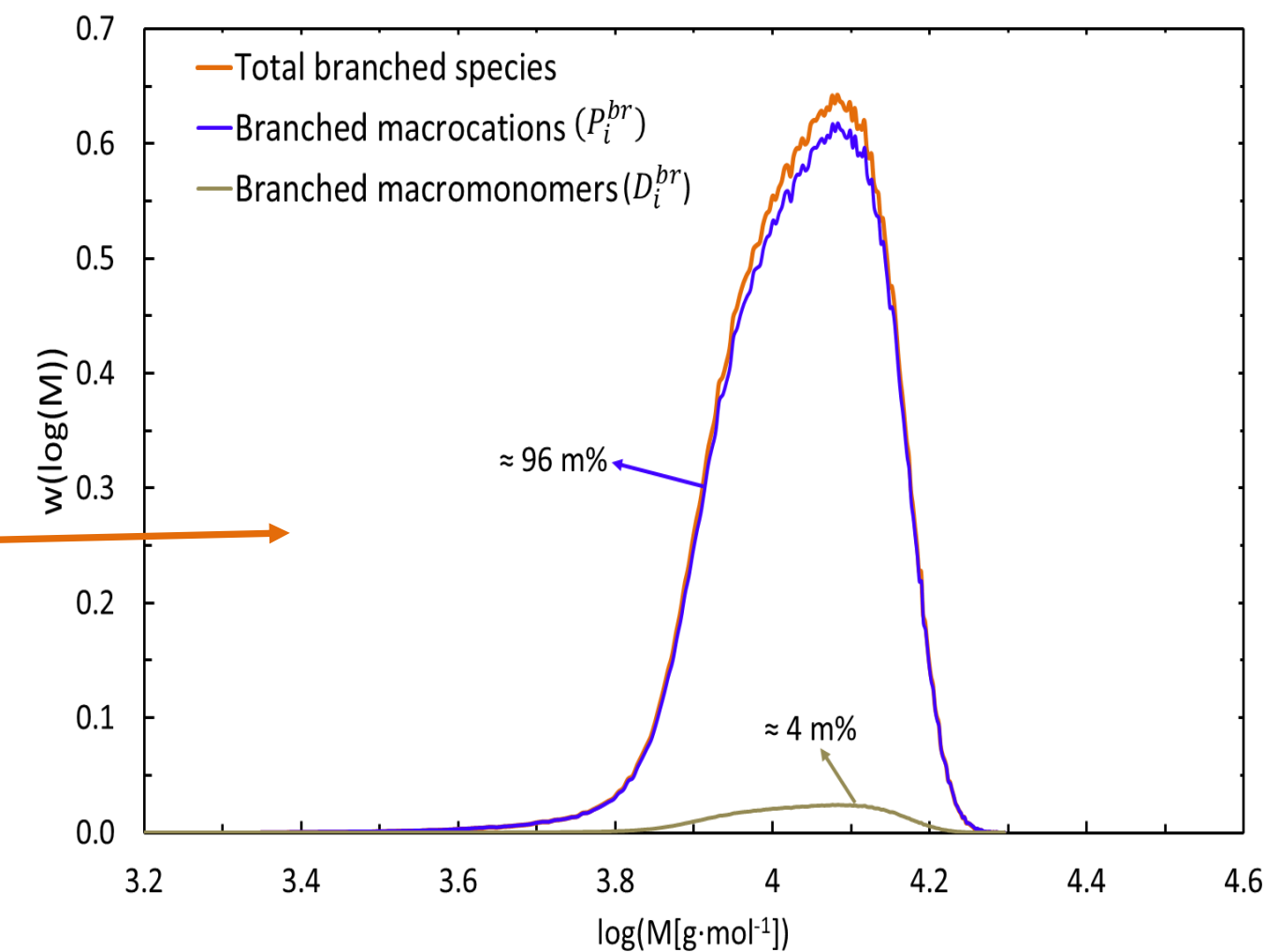
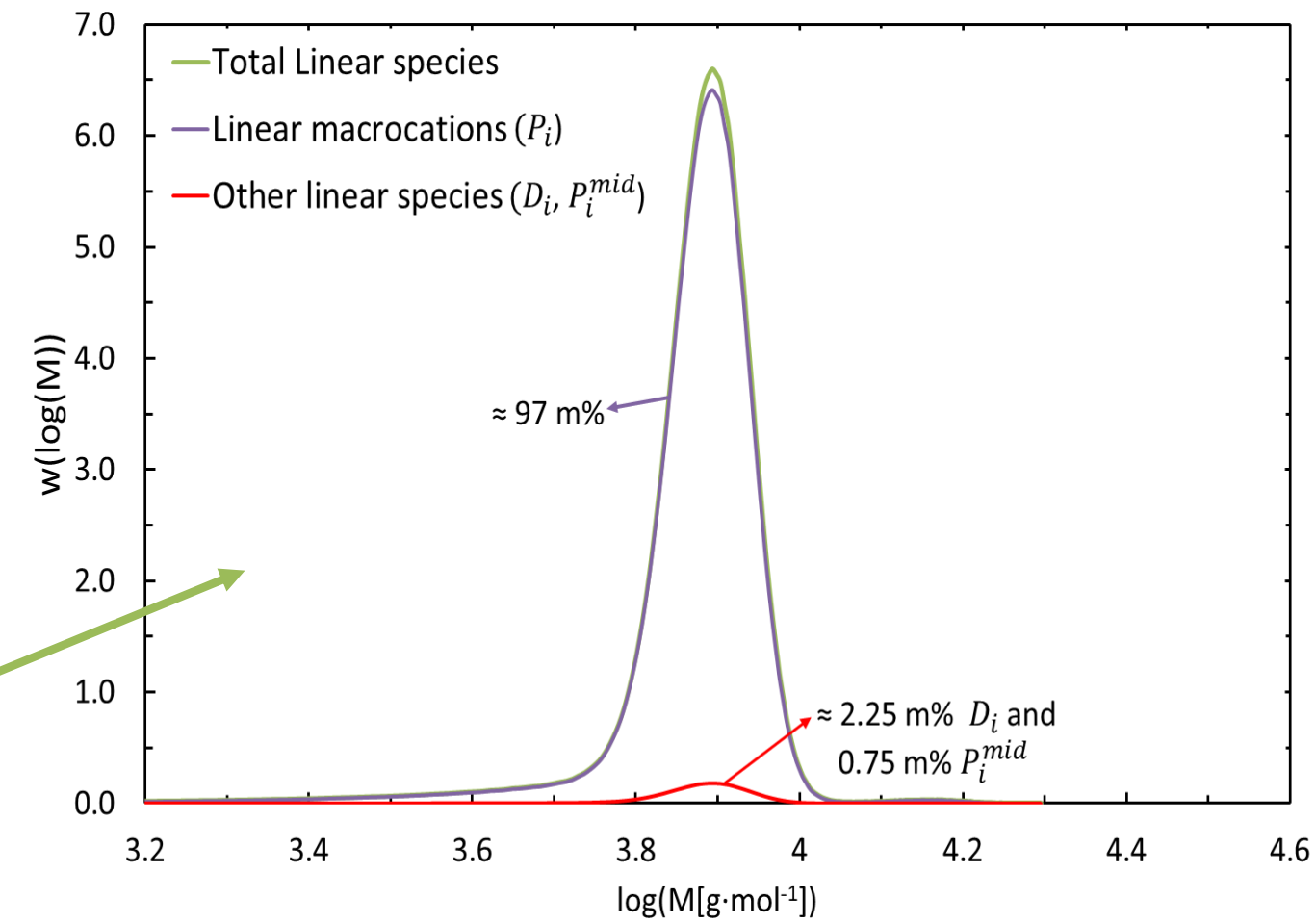
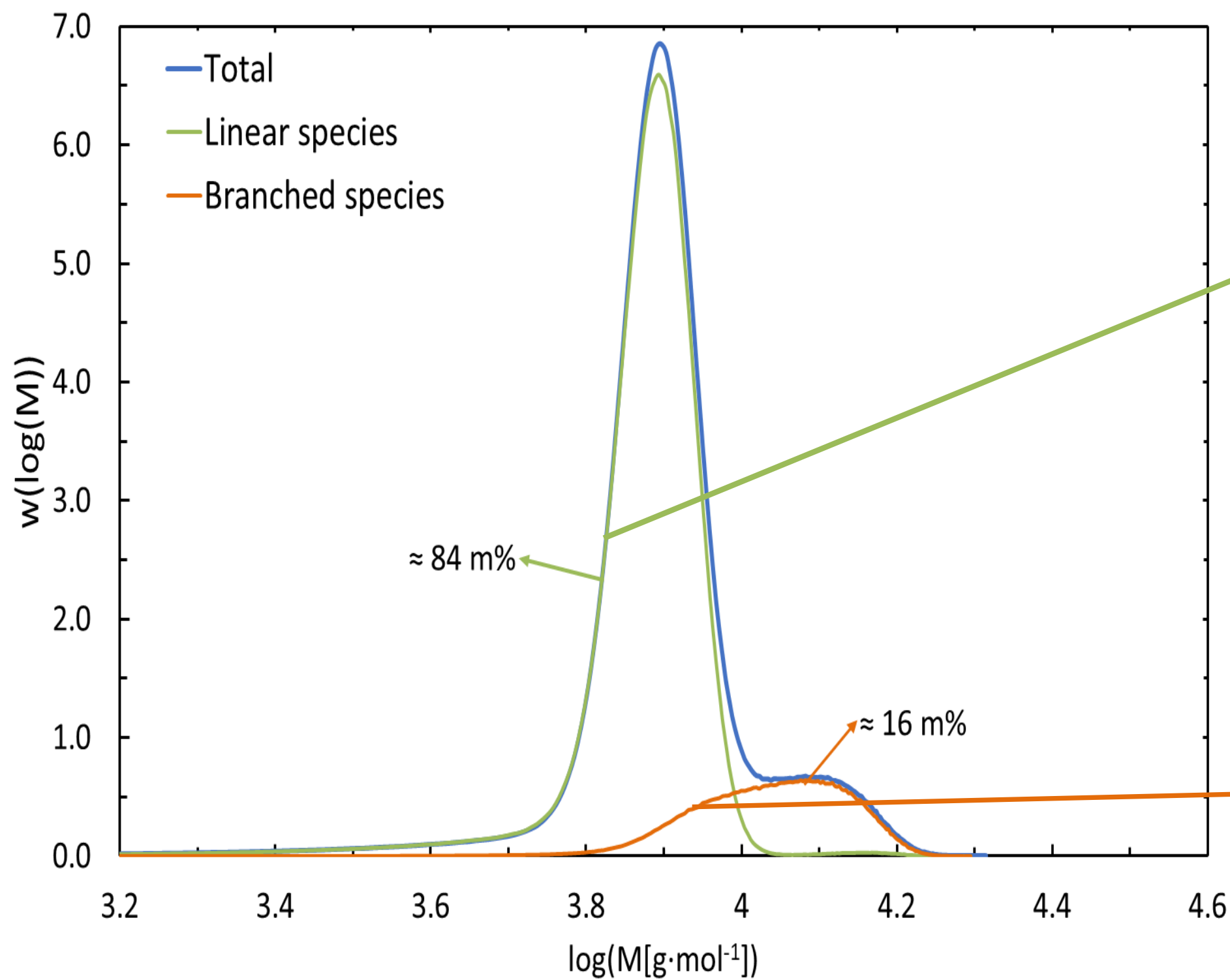
	$k_{trM}$ (L·mol <sup>-1</sup> ·s <sup>-1</sup> )	$k_{pm}$ (L·mol <sup>-1</sup> ·s <sup>-1</sup> )	$k_{pMid}$ (L·mol <sup>-1</sup> ·s <sup>-1</sup> )
1. —	✗	✗	✗
2. —	✓ 6.57×10 <sup>-2</sup>	✗	✗
3. —	✓ 6.57×10 <sup>-2</sup>	✓ 6.99×10 <sup>-2</sup>	✗
4. —	✓ 6.57×10 <sup>-2</sup>	✓ 6.99×10 <sup>-2</sup> ×0.1	✓ 3.49×10 <sup>-2</sup>
5. —	✓ 6.57×10 <sup>-2</sup>	✓ 6.99×10 <sup>-3</sup>	✓ 3.49×10 <sup>-2</sup> ×10
6. - -	✓ 6.57×10 <sup>-2</sup>	✓ 6.99×10 <sup>-3</sup>	✓ 3.49×10 <sup>-1</sup>
● - -		Experimental	



Correction for SEC broadening and topology difference

# EXTRA INFORMATION BY MODELING

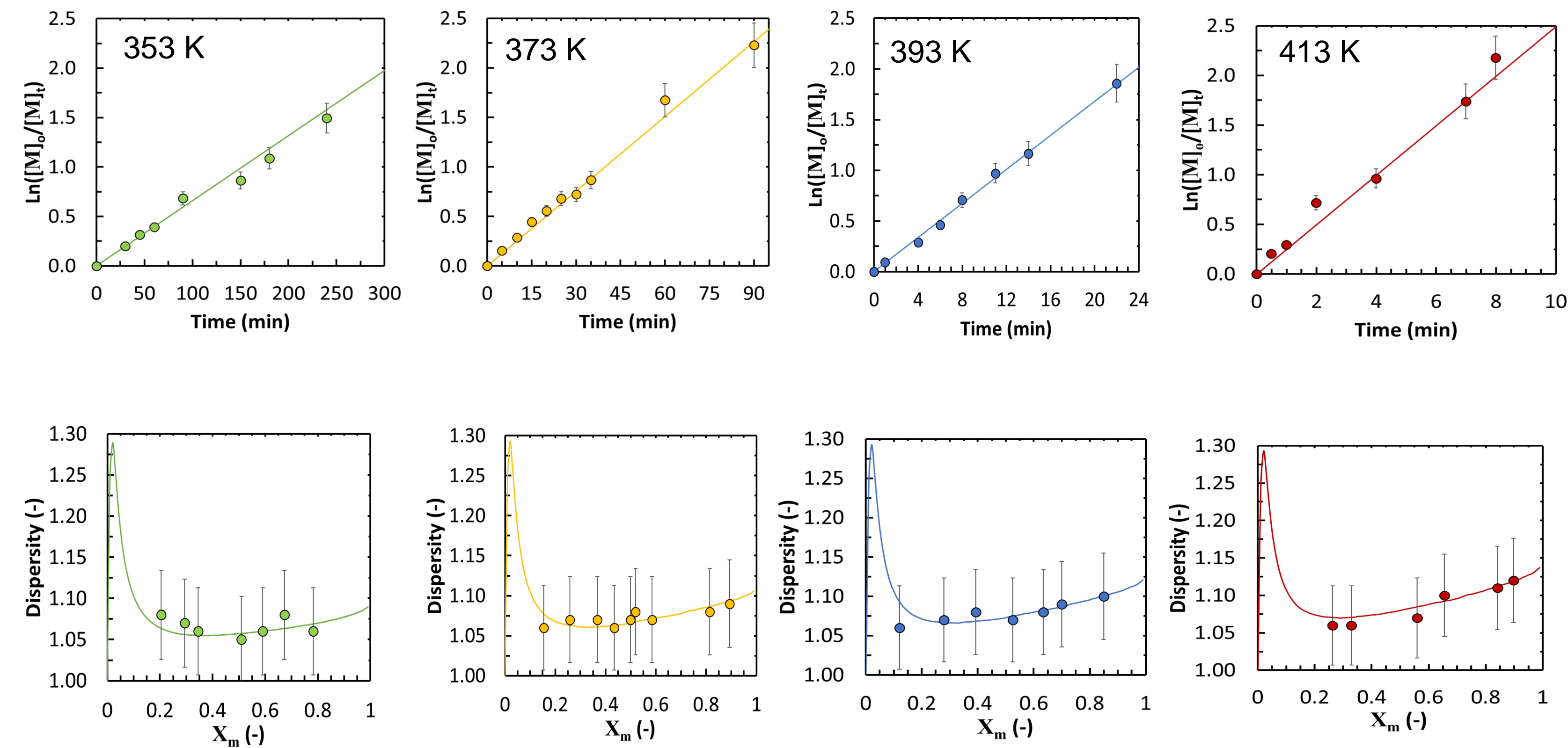
*Differentiation according to topology*



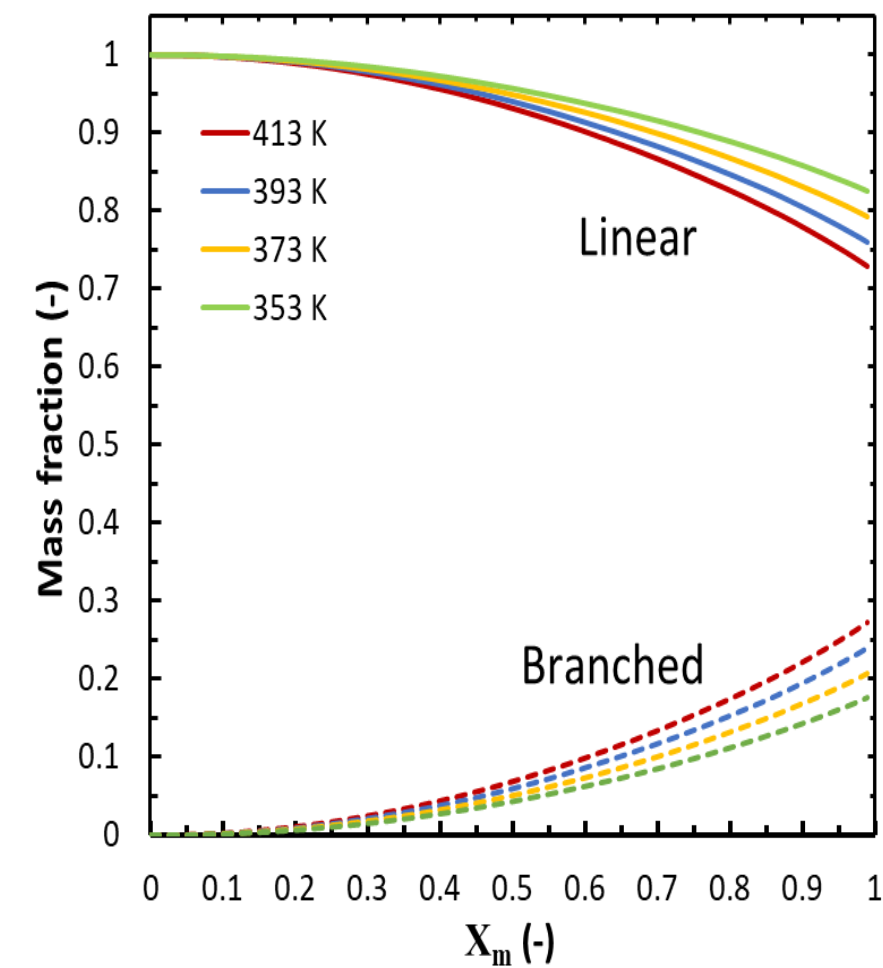
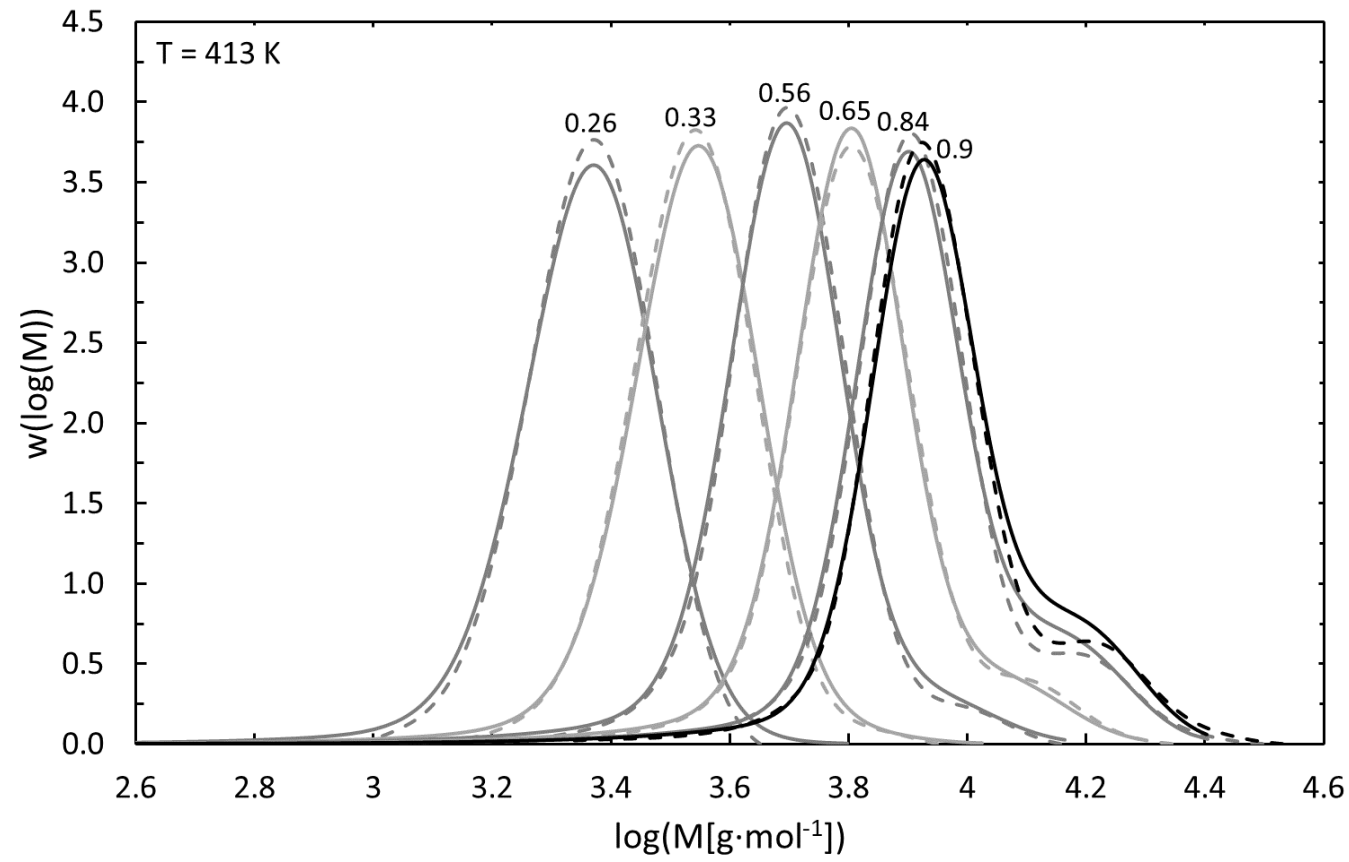
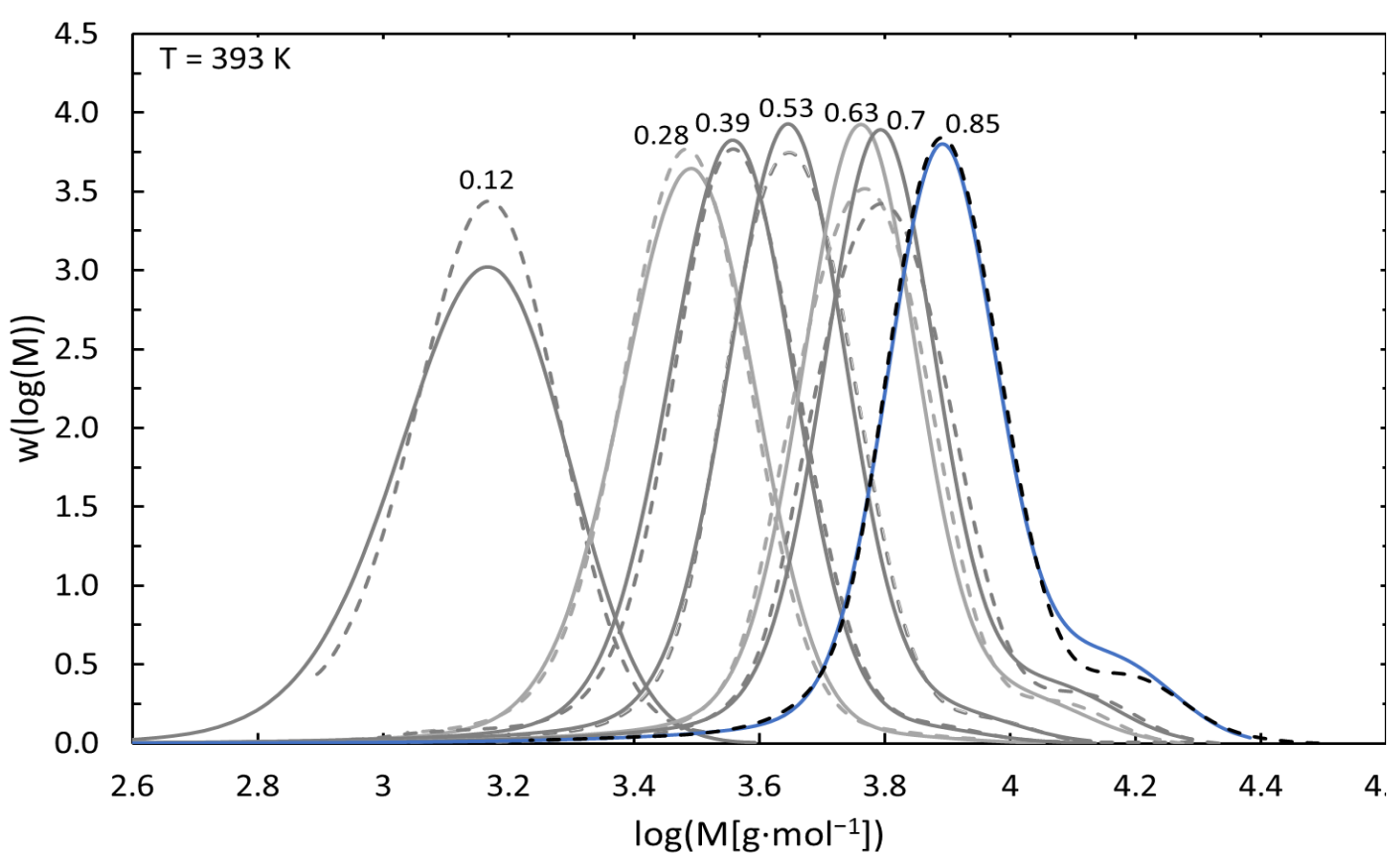
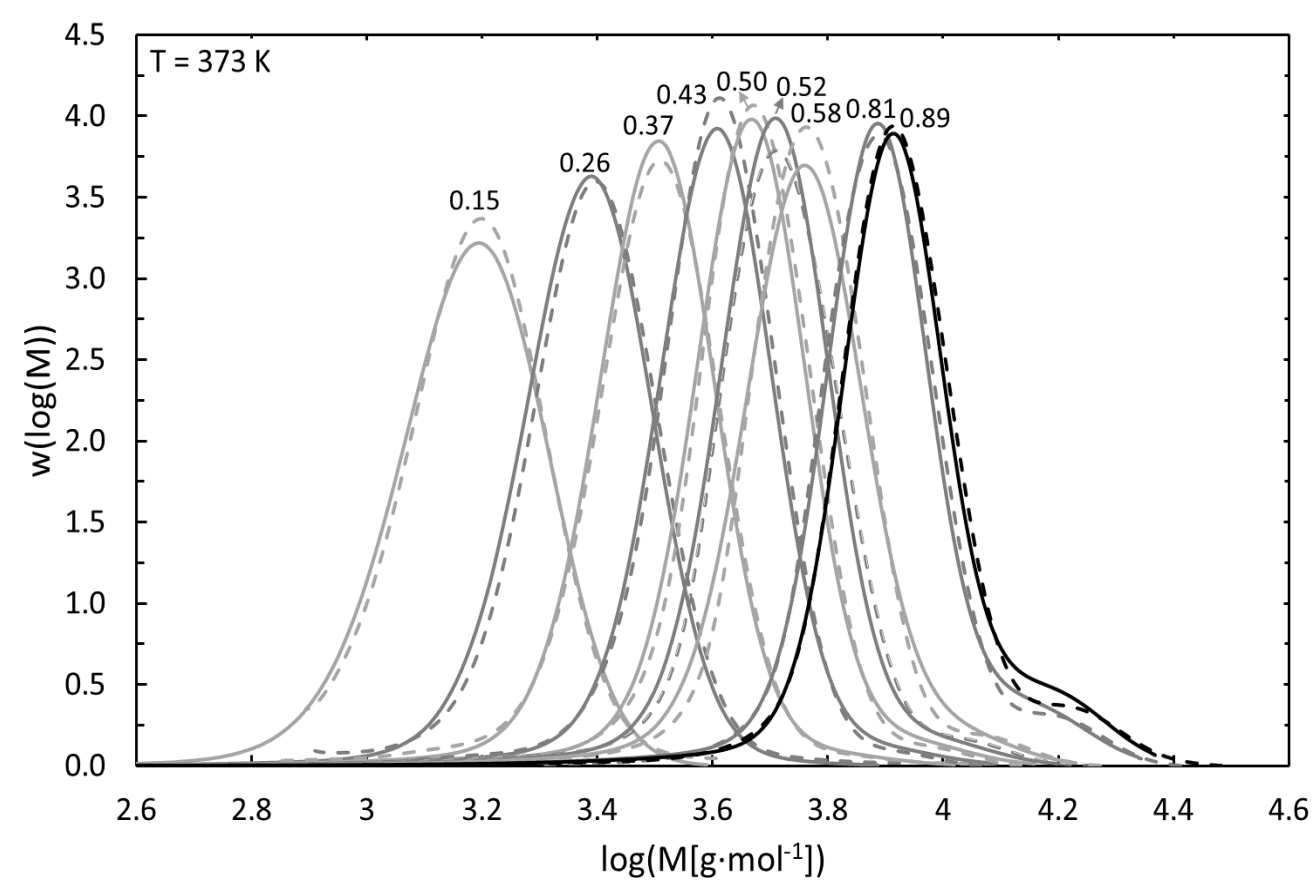
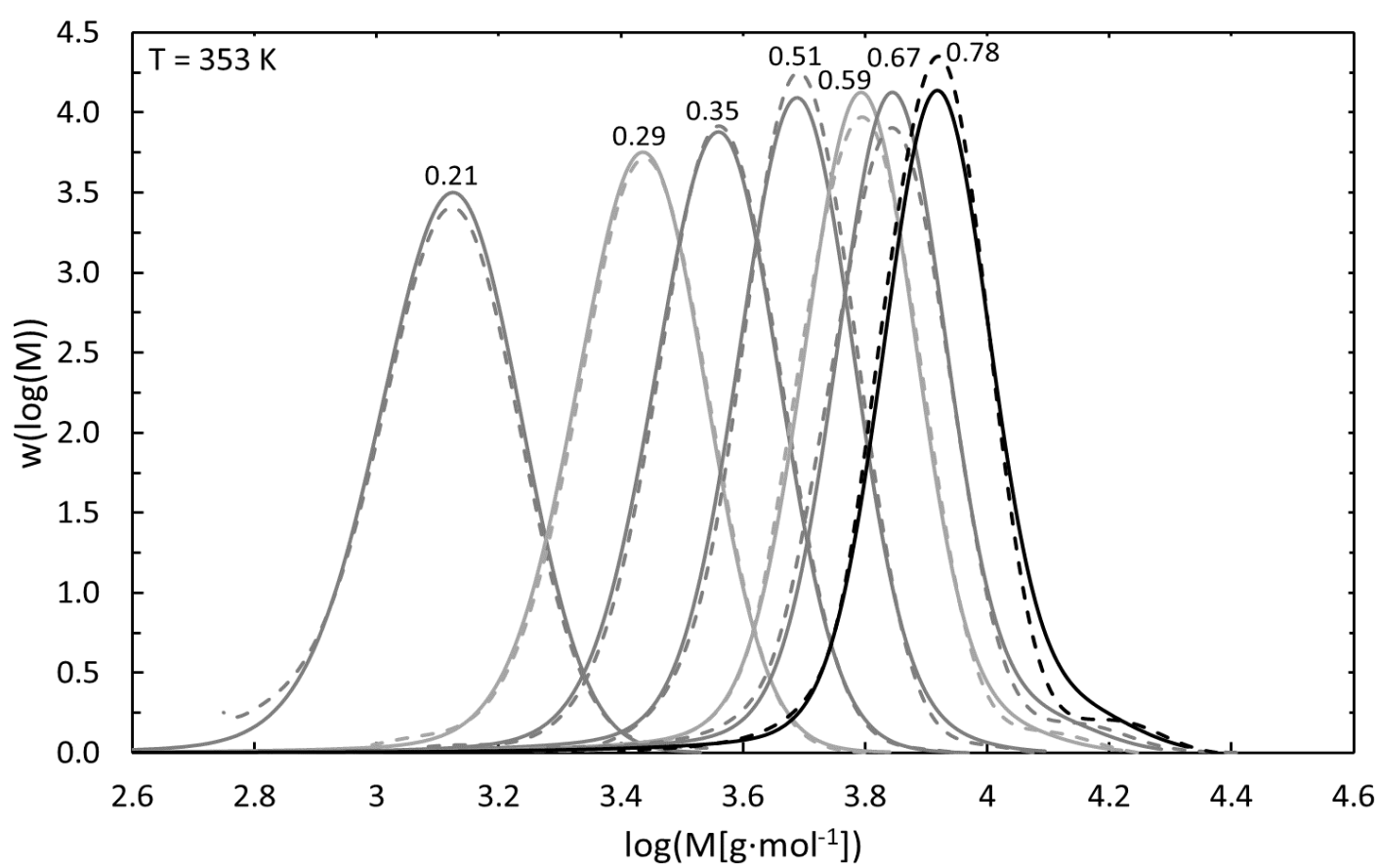
*Absolute SEC traces accessible*



# SAME PROCEDURE (T) TO OBTAIN ARRHENIUS PARAMETERS



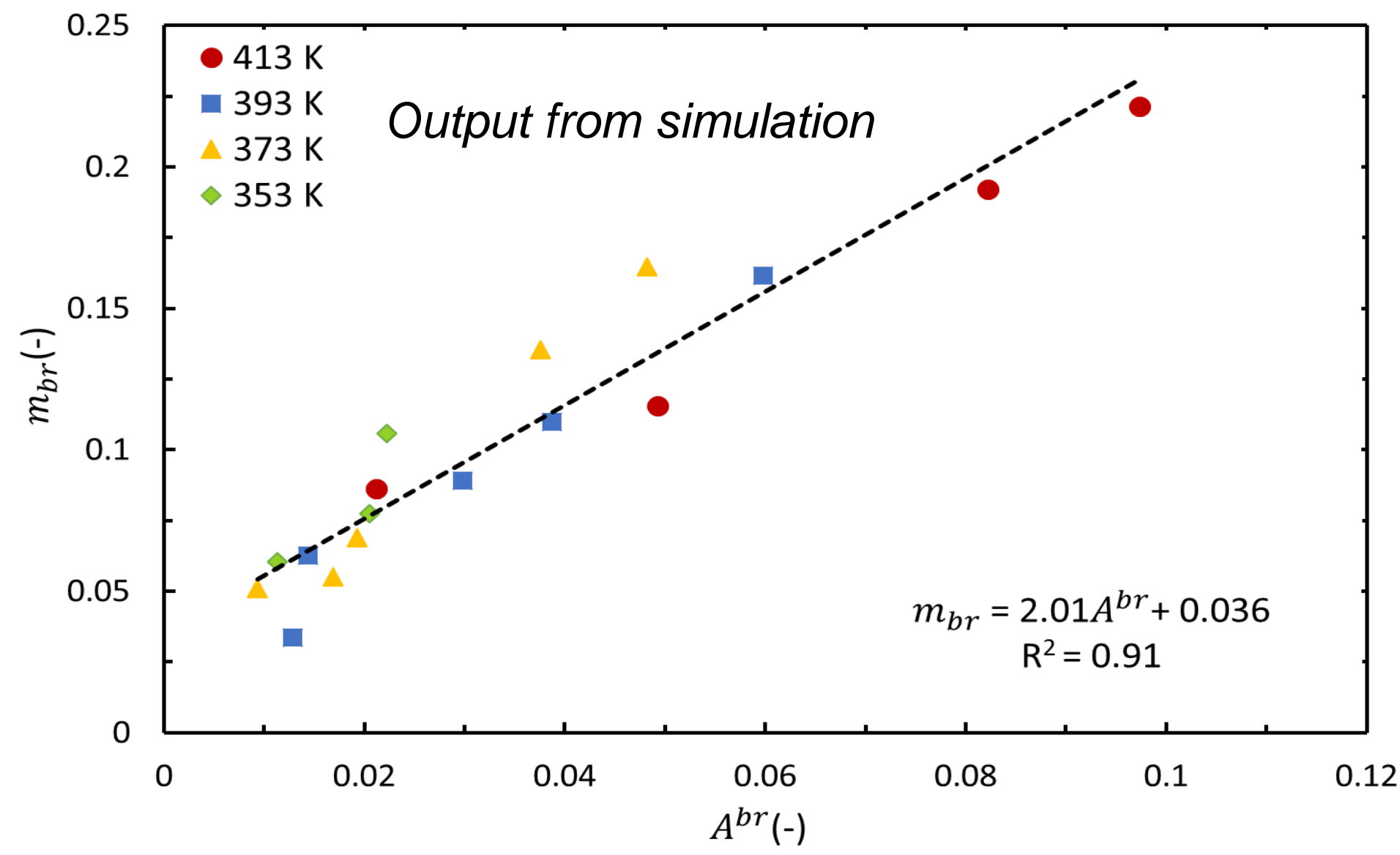
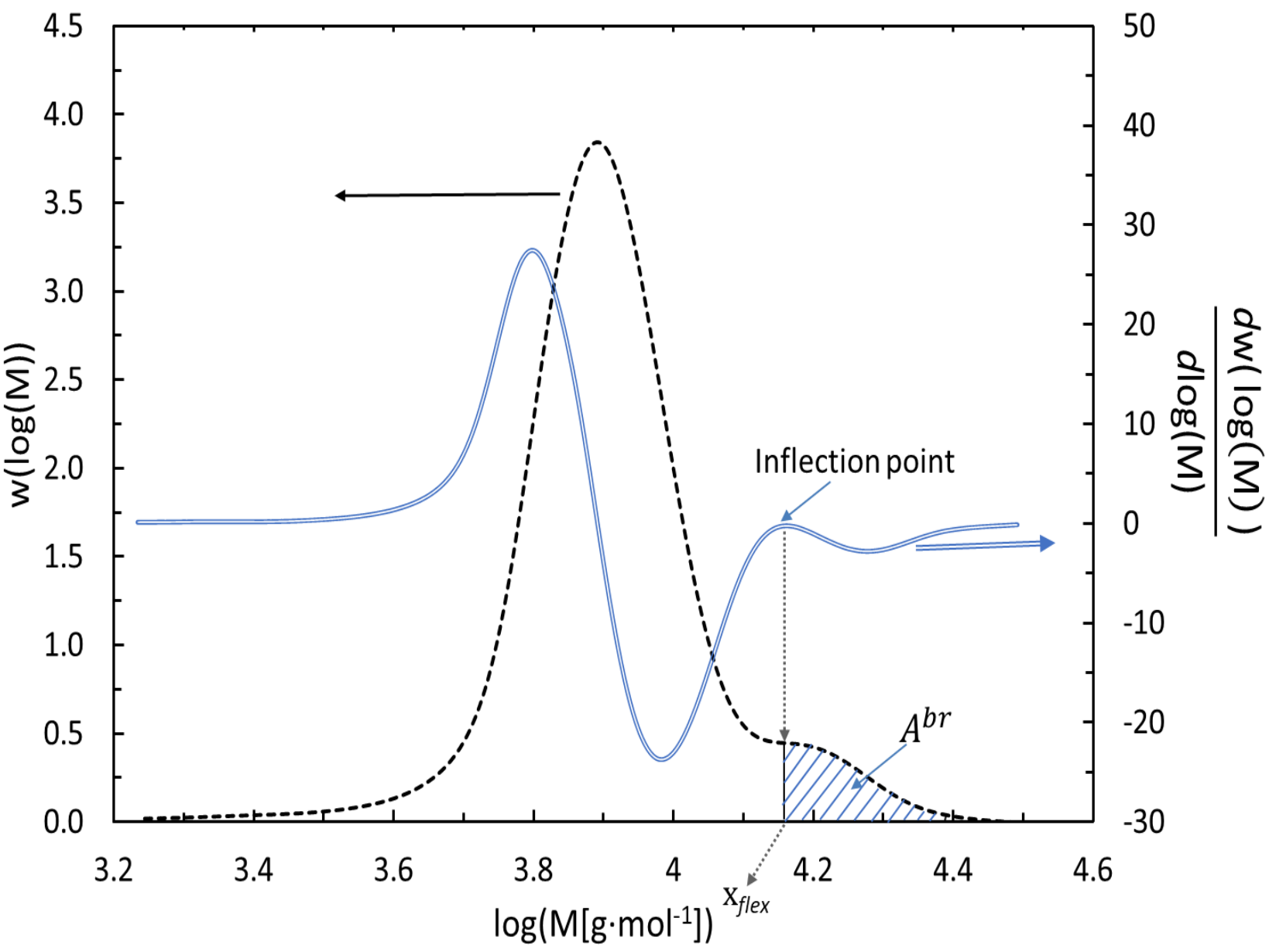
# SAME PROCEDURE (T) TO OBTAIN ARRHENIUS PARAMETERS





# REVERSE ENGINEERING: DIRECT EXPERIMENTAL ANALYSIS ?

*Linear calibration curve to assess the importance of branched species*



*Input from experimental SEC trace*

# OUTLINE

## 1. Homopolymerization

- Chain initiation and propagation reactivity
- Chain transfer to monomer reactivity
- Macropropagation reactivity

Van Steenberge *et al. Macromolecules* **2015**, 48, 7765  
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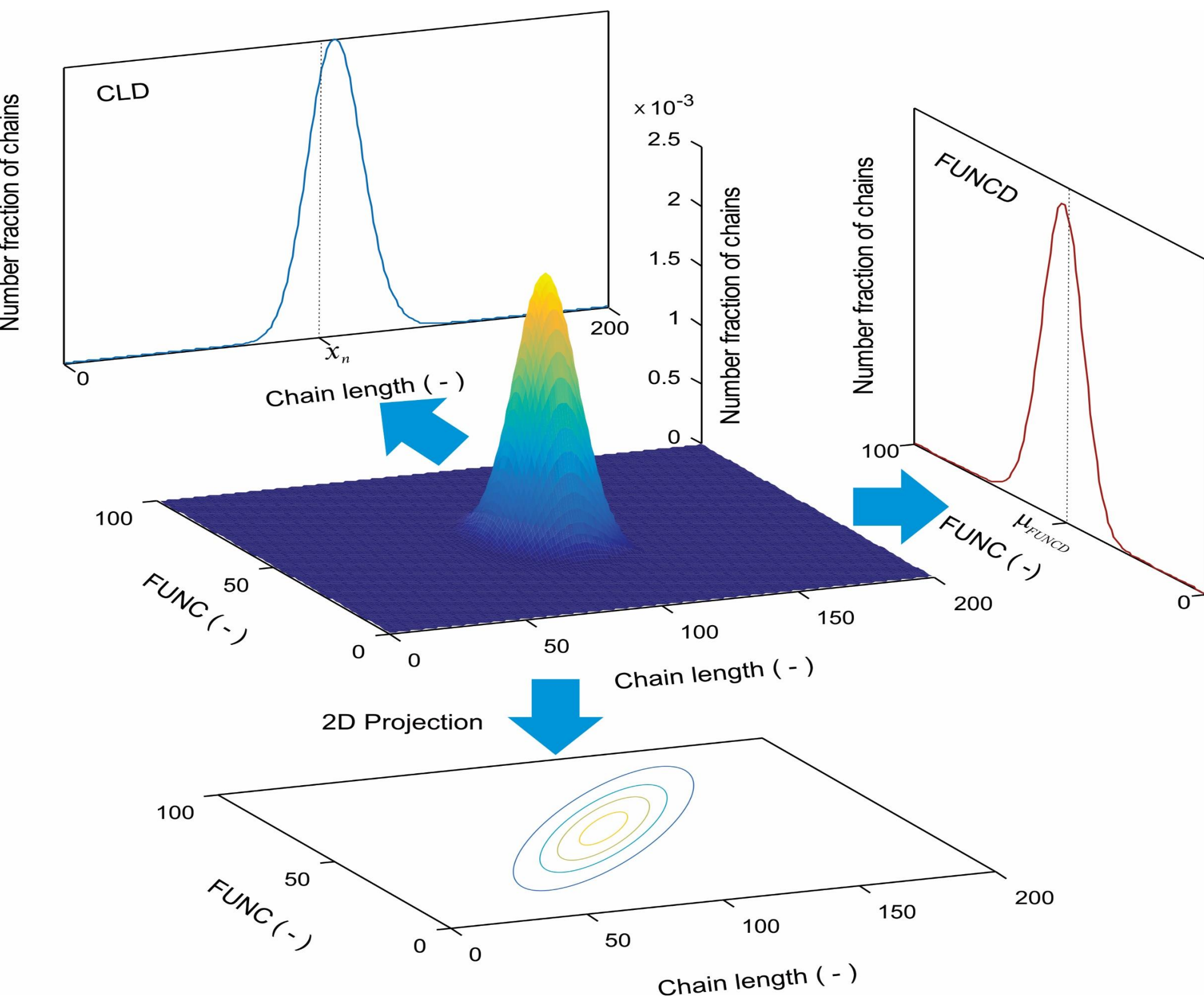
## 2. Copolymerization with functional comonomer in view of therapeutics & hydrogels

- Unbiased qualification tool for evaluation functionalization success
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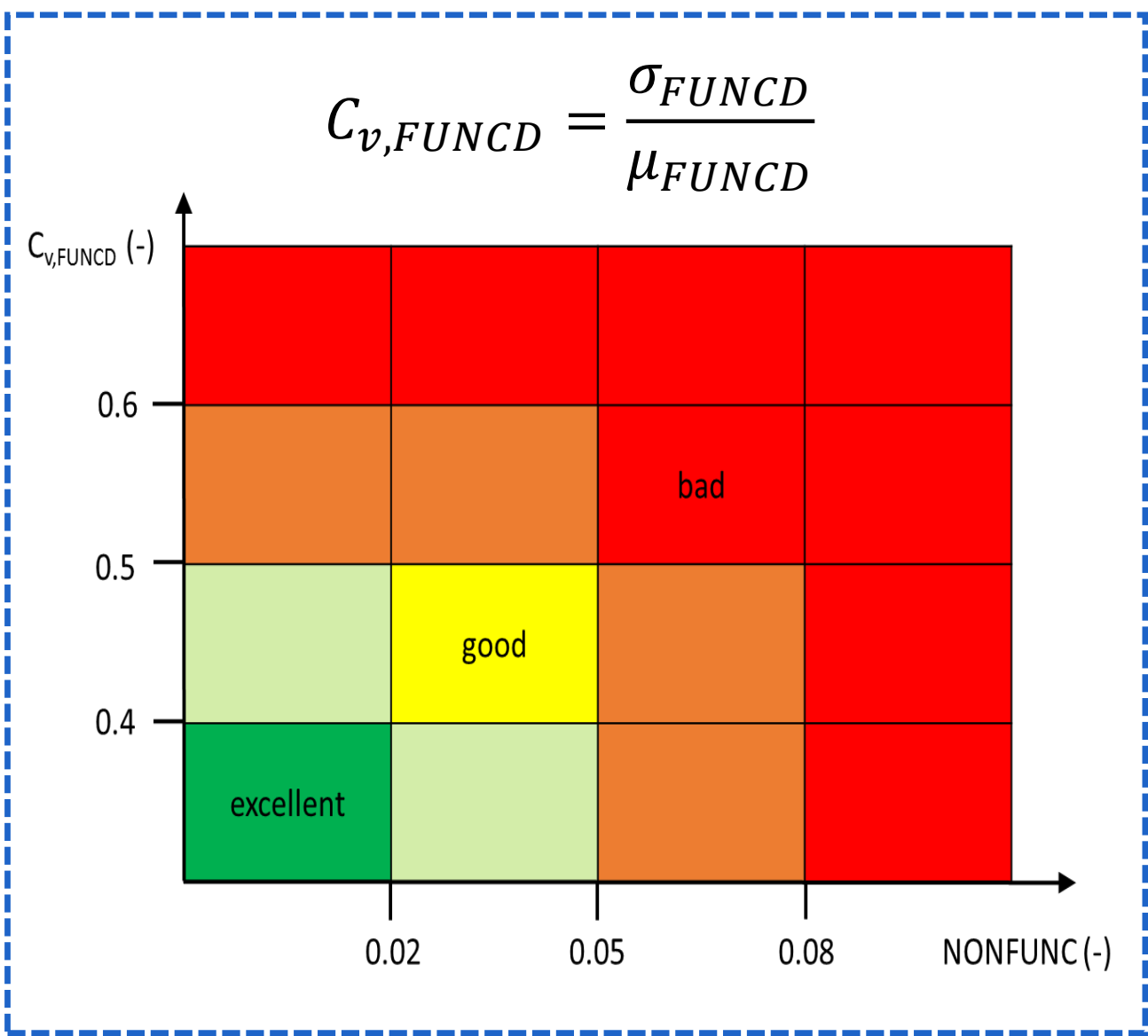
Van Steenberge *et al. Macromolecules* **2015**, 48, 7765  
Van Steenberge *et al. Nat. Commun.* **2019**, to be subm.



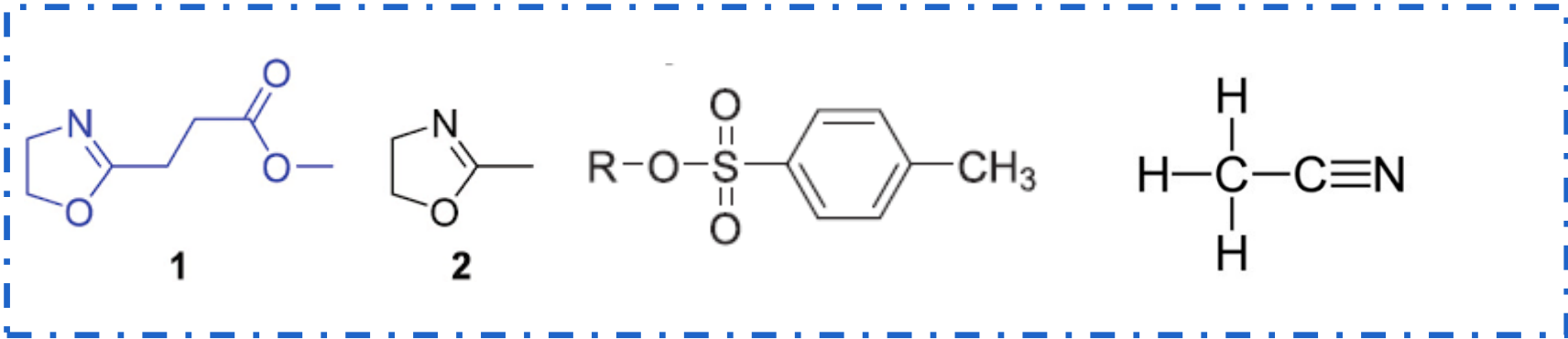
# CONCEPT OF FUNCTIONALITY-CHAIN LENGTH DISTRIBUTION



*derived properties*

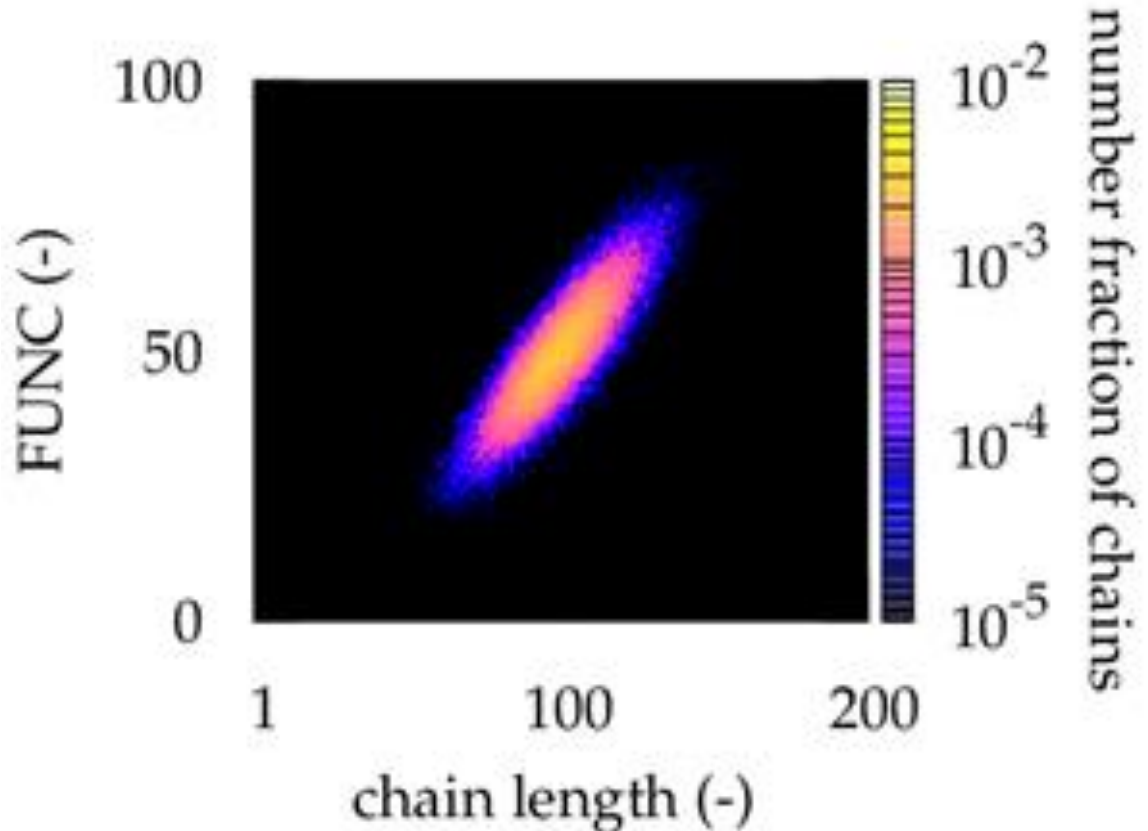


# APPLICATION FOR CROP OF MEOX AND C2MESTOX: 1. EQUIMOLAR

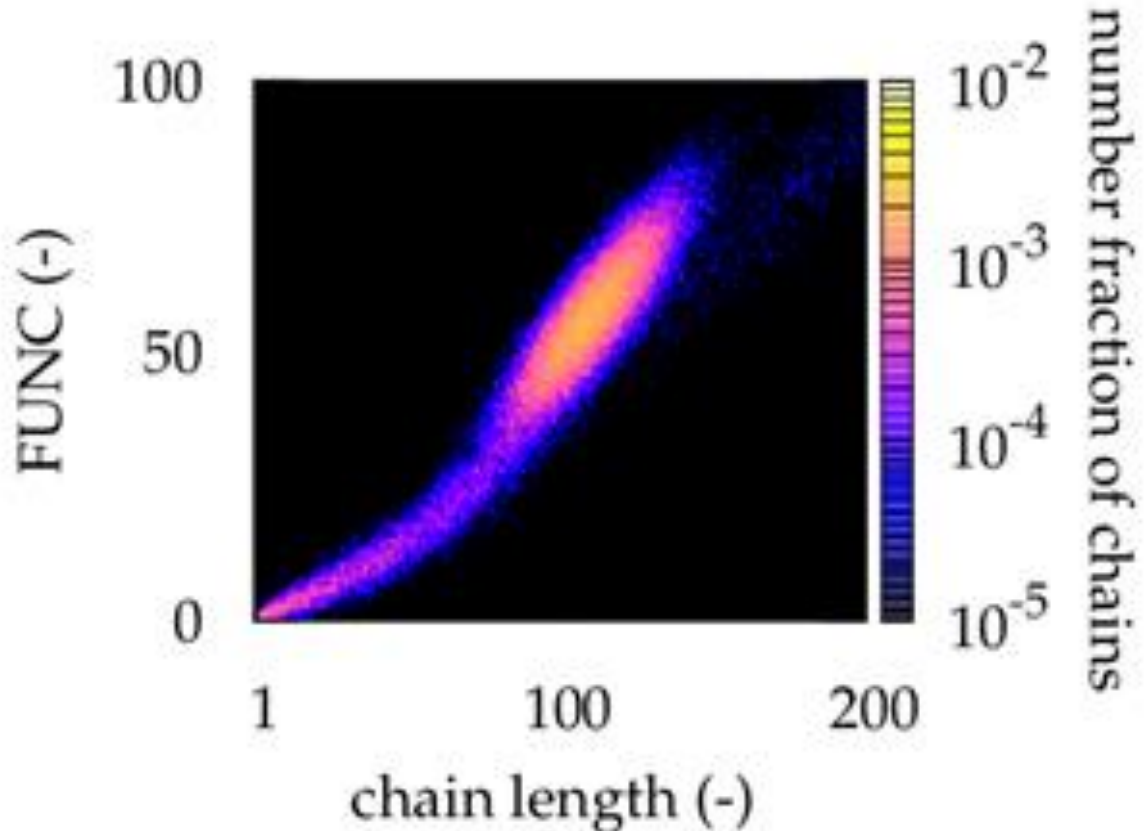


*fingerprint of copolymer product quality*  
*rate coefficients cf. first part*

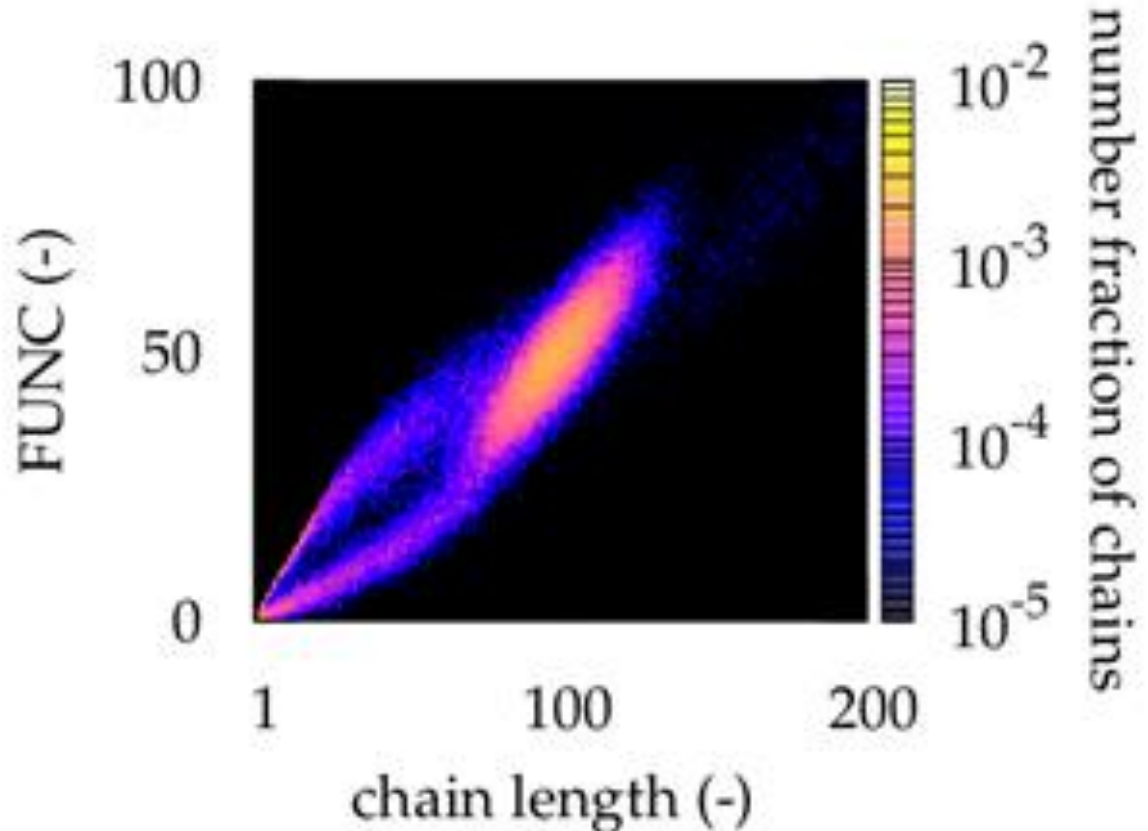
no side reactions



chain transfer to monomer



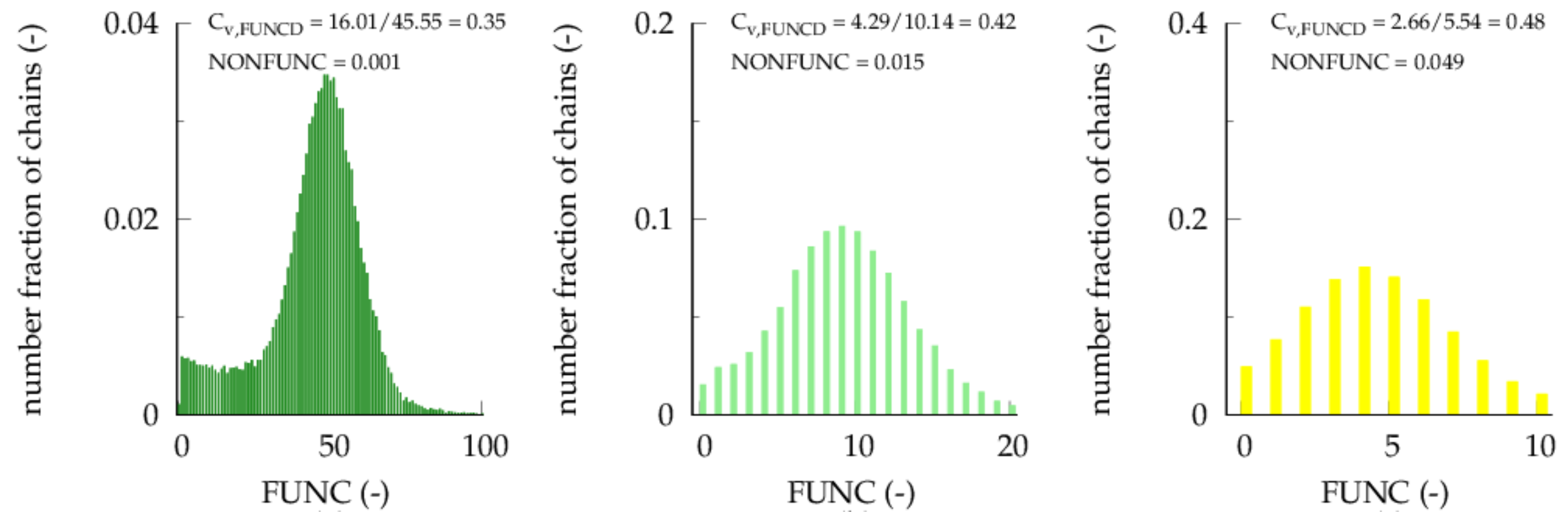
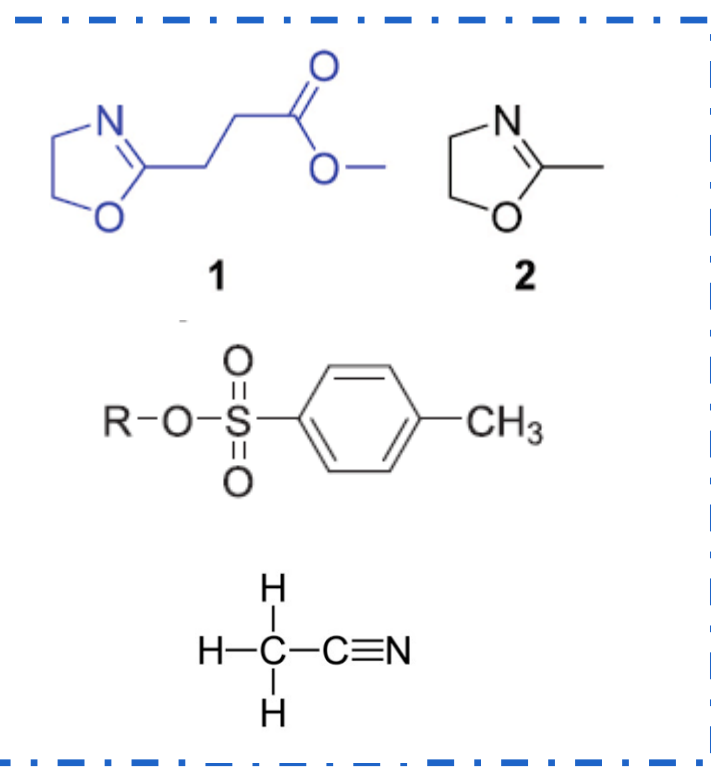
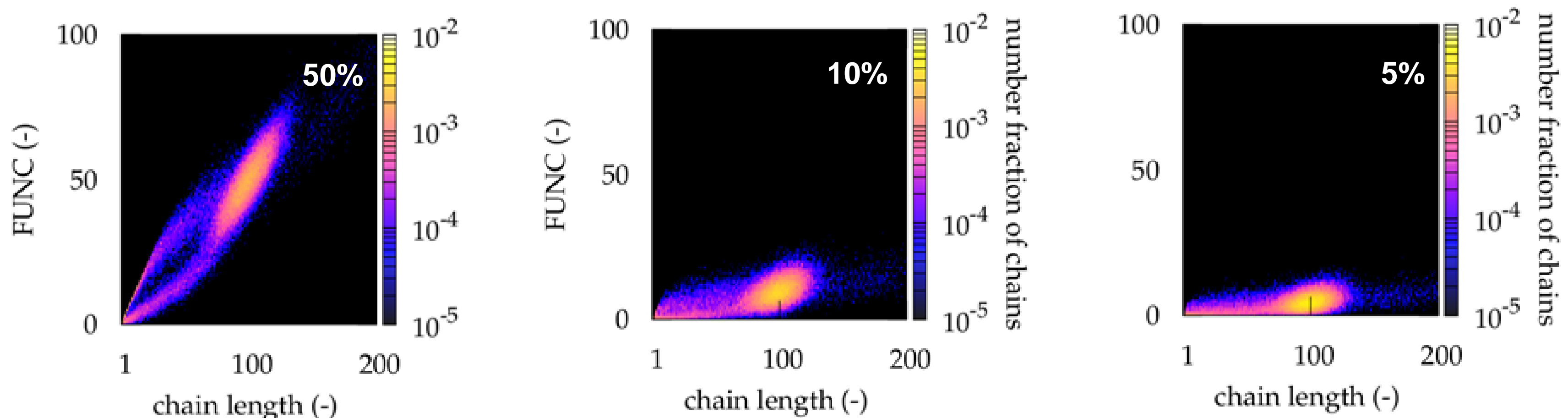
+ extra chain initiation  
& macropropagation



*total monomer concentration: 3 mol L<sup>-1</sup>; solvent acetonitrile; target DP of 100; 413 K;*  
*overall monomer conversion of 100%*



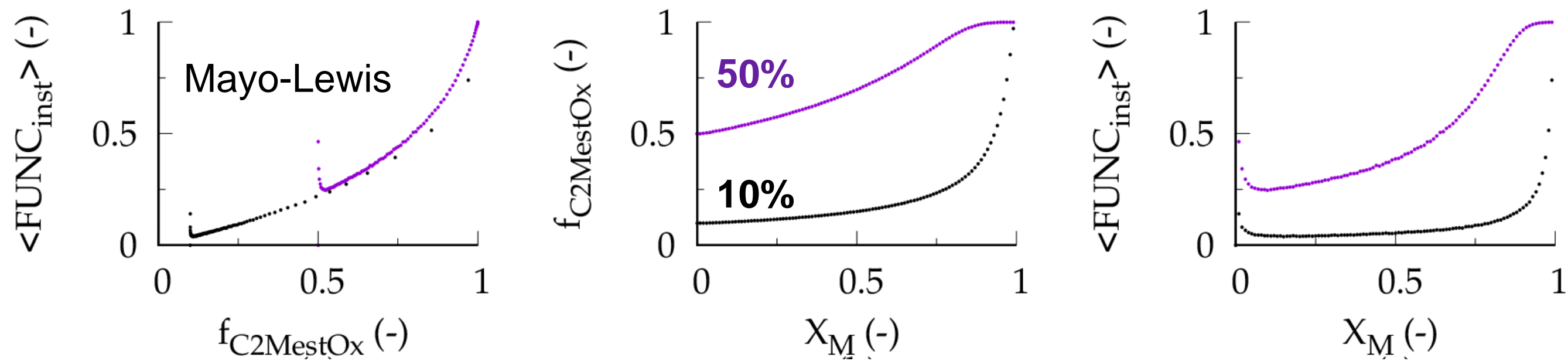
# APPLICATION FOR CROP OF MEOX AND C2MESTOX: 2. LOWER AMOUNTS



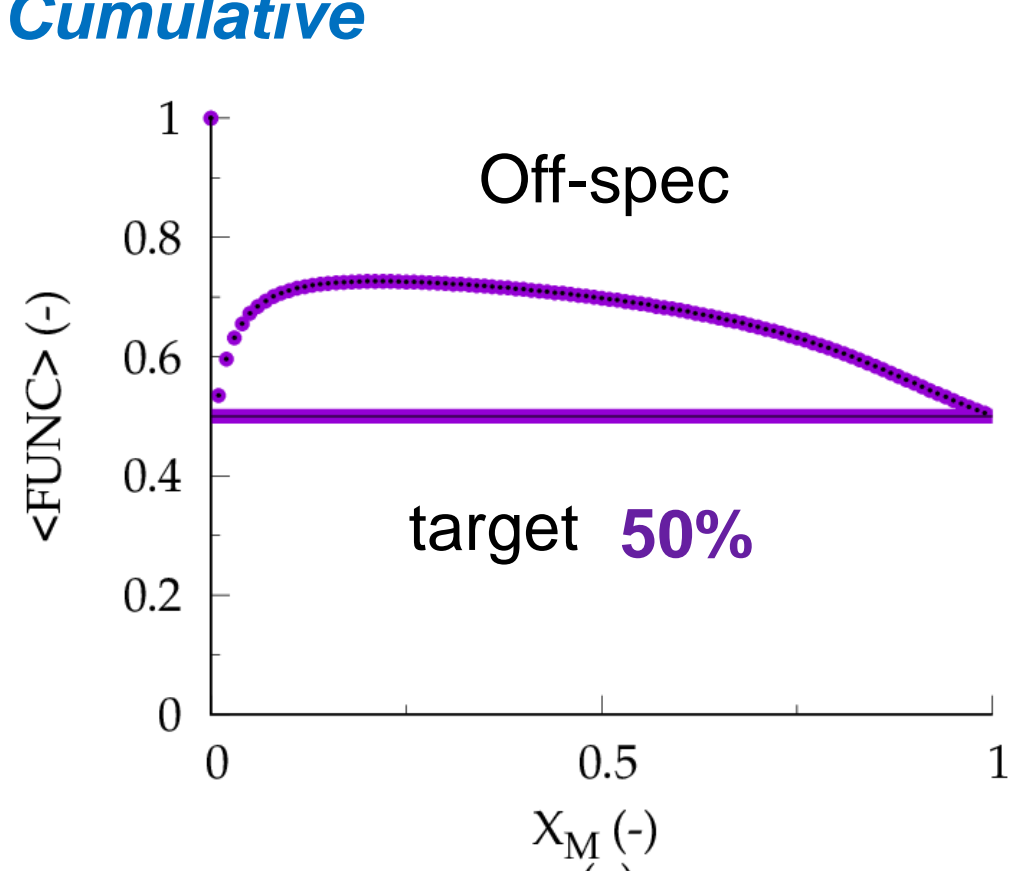
# APPLICATION FOR CROP OF MEOX AND C2MESTOX

*Unbiased in contrast  
to conventional average  
characterization*

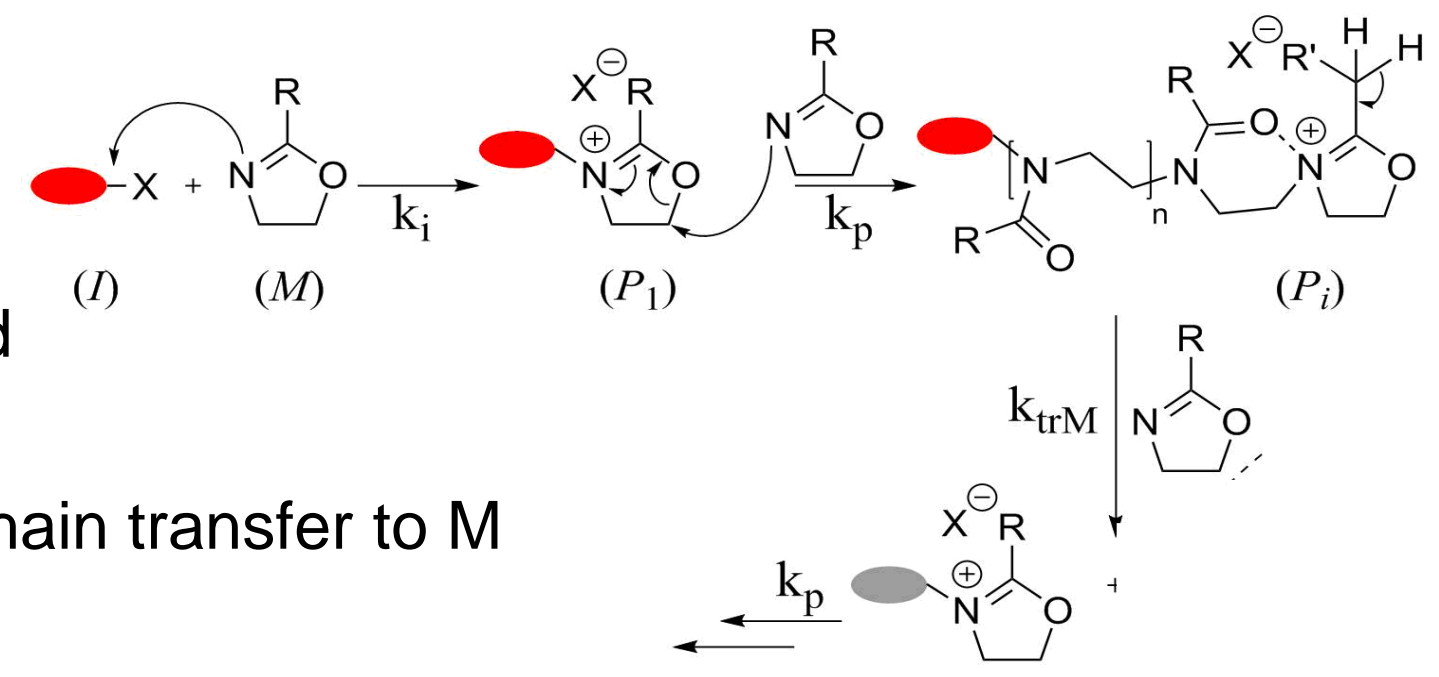
*Instantaneous*



*Cumulative*

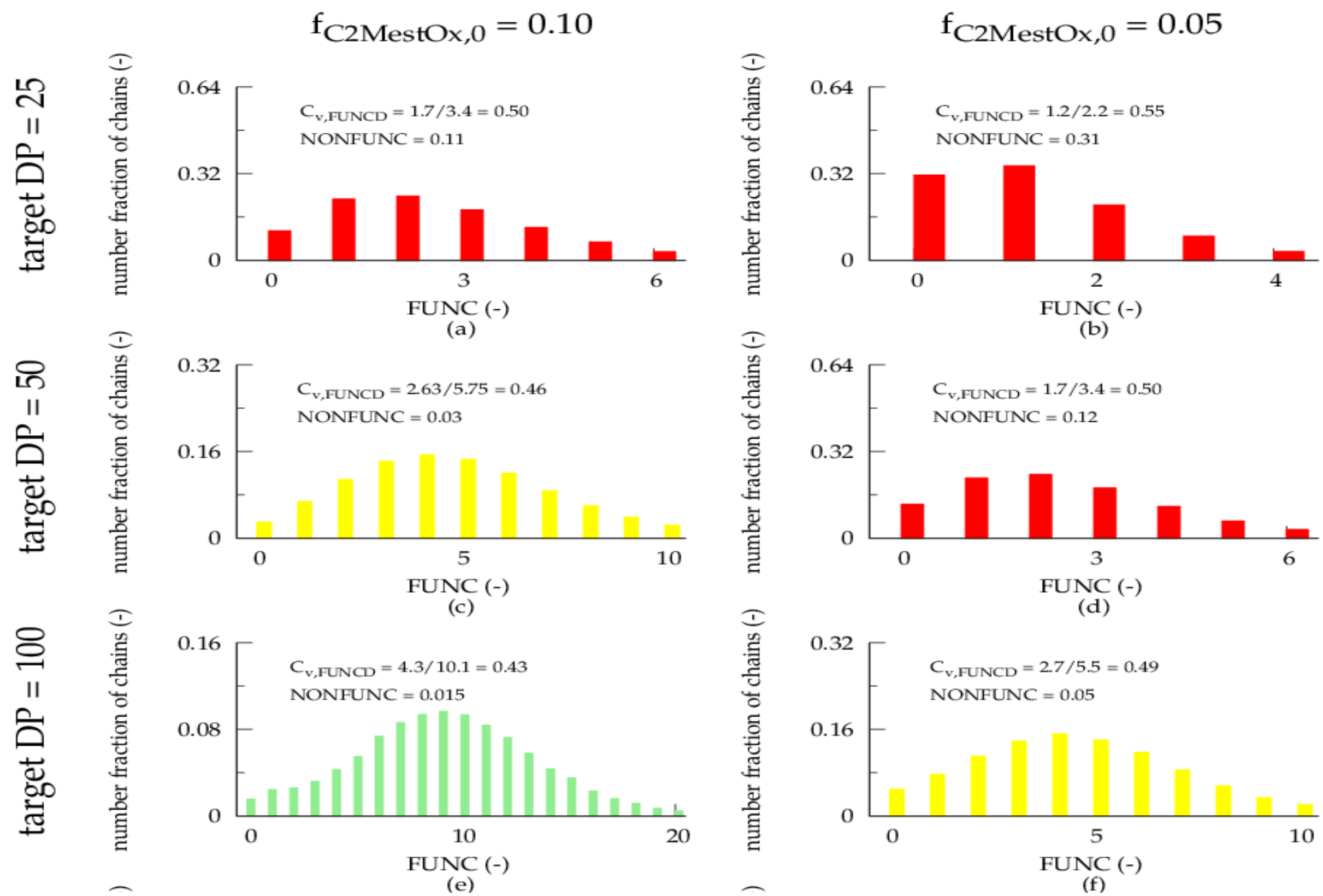


Chain length history not included  
**side reactions masked**  
Same results with and without chain transfer to M



# STRENGTH OF MODEL: FULL SCREENING OF CONDITIONS

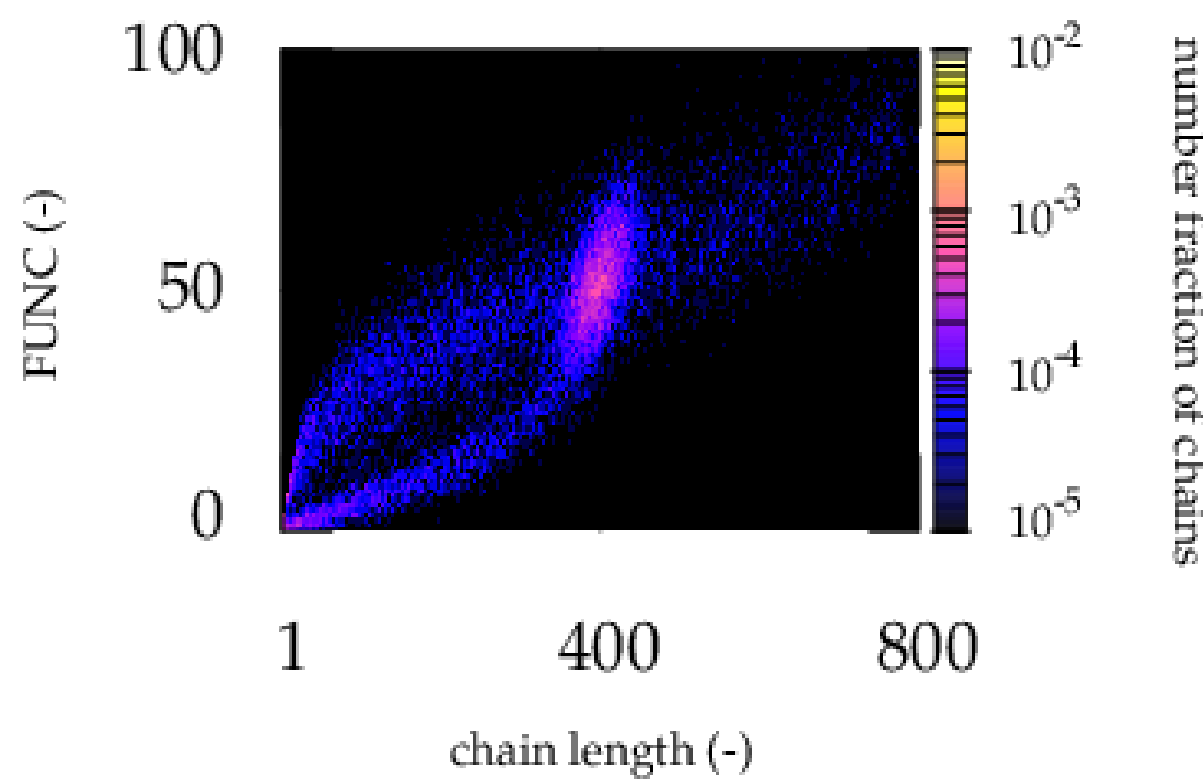
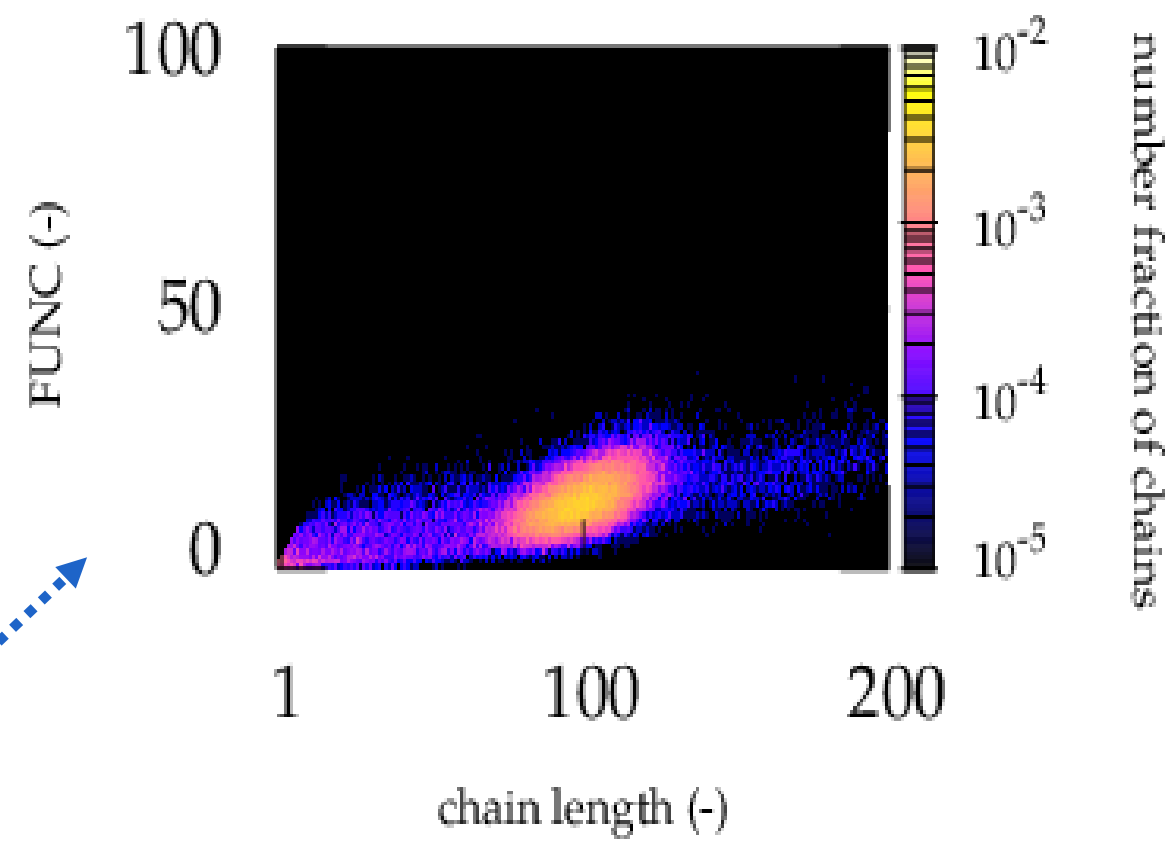
e.g. 6 cases



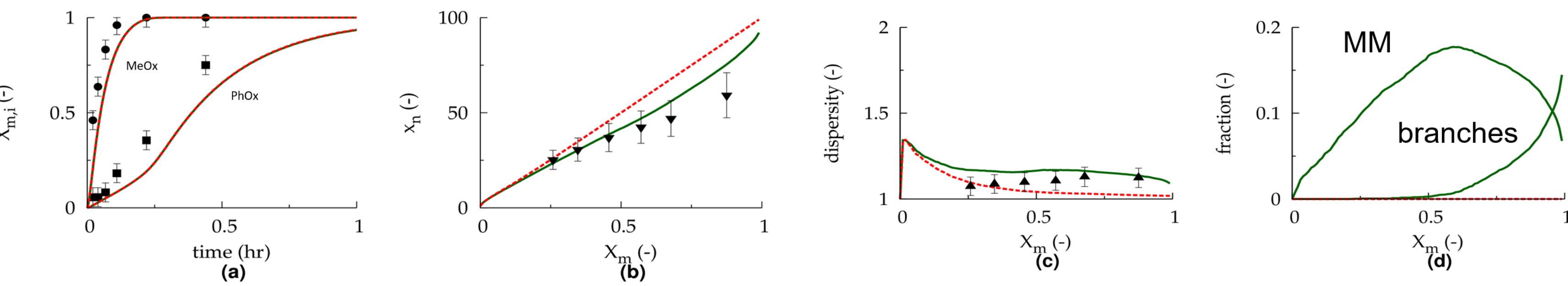
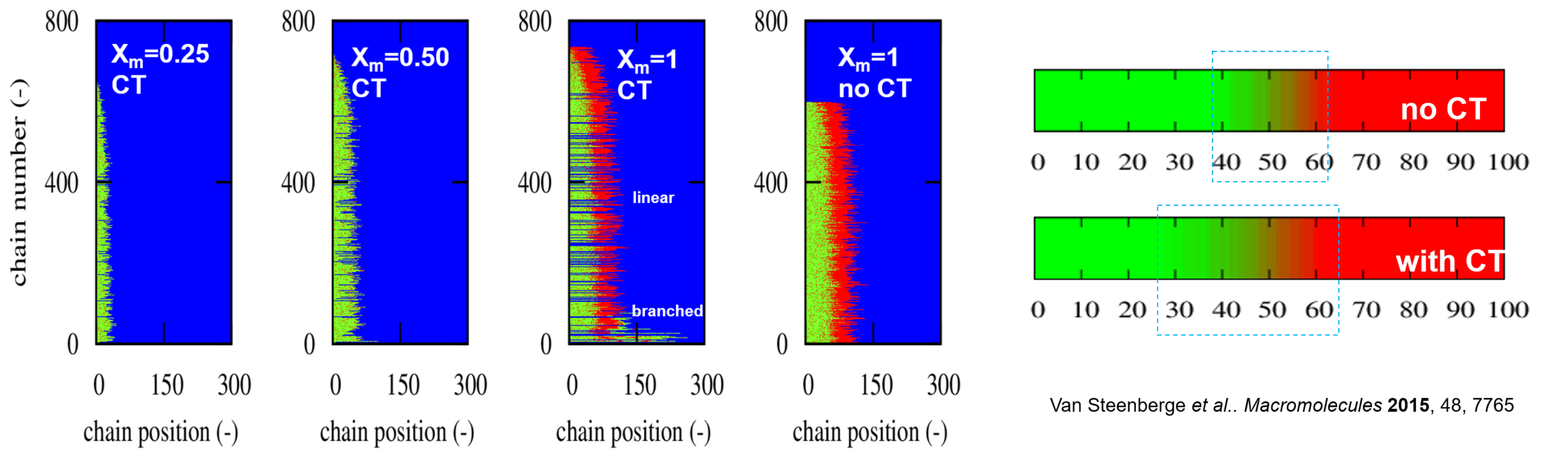


# STRENGTH OF MODEL: DESIGN OF CROP PROCESS

Reaction conditions		$C_{v,FUNCD}$ & NONFUNC (-)	
		MeOx/C2MestOx	EtOx/C2MestOx
1	$f_{C2MestOx,0}=0.02$ ; target DP=100; T=100°C	0.527 & 0.195	0.522 & 0.213
2	$f_{C2MestOx,0}=0.02$ ; target DP=100; T=140°C	0.531 & 0.205	0.539 & 0.238
3	$f_{C2MestOx,0}=0.02$ ; target DP=400; T=100°C	0.554 & 0.072	0.627 & 0.106
4	$f_{C2MestOx,0}=0.02$ ; target DP=400; T=140°C	0.602 & 0.101	0.677 & 0.137
5	$f_{C2MestOx,0}=0.10$ ; target DP=100; T=100°C	0.415 & 0.012	0.413 & 0.019
6	$f_{C2MestOx,0}=0.10$ ; target DP=100; T=140°C	0.424 & 0.015	0.447 & 0.026
7	$f_{C2MestOx,0}=0.10$ ; target DP=400; T=100°C	0.497 & 0.011	0.599 & 0.014
8	$f_{C2MestOx,0}=0.10$ ; target DP=400; T=140°C	0.564 & 0.018	0.672 & 0.021
9	<b><math>f_{C2MestOx,0}=0.13</math>; target DP=100; T=100°C</b>	<b>0.393 &amp; 0.008</b>	0.396 & 0.011
10	$f_{C2MestOx,0}=0.13$ ; target DP=100; T=140°C	0.411 & 0.012	0.432 & 0.016
11	<b><math>f_{C2MestOx,0}=0.13</math>; target DP=400; T=100°C</b>	<b>0.481 &amp; 0.008</b>	0.593 & 0.009
12	$f_{C2MestOx,0}=0.13$ ; target DP=400; T=140°C	0.561 & 0.013	0.664 & 0.013

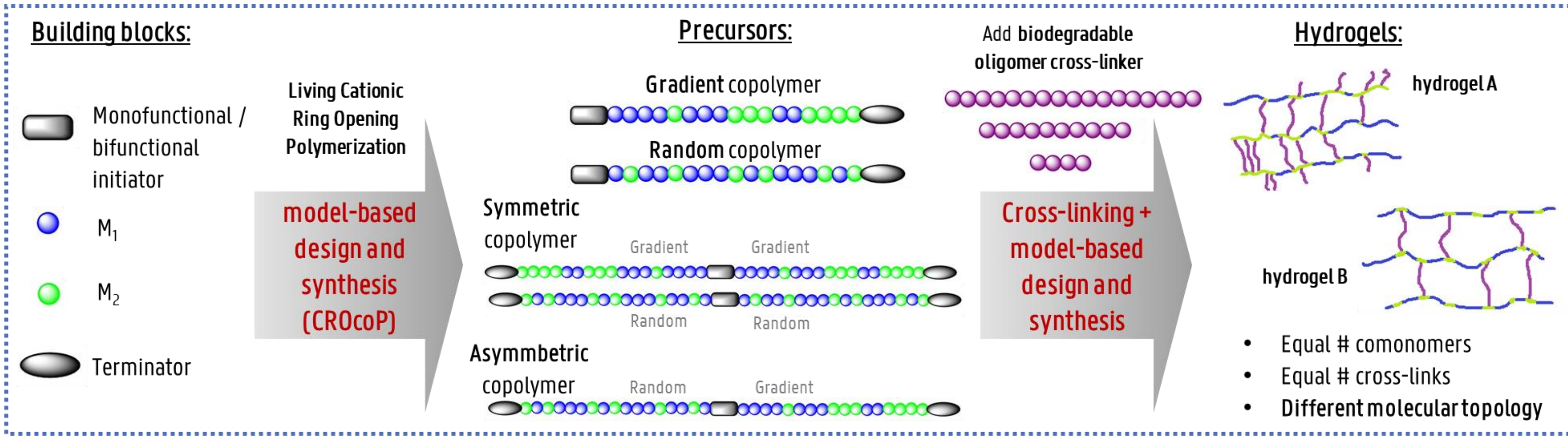


# NEXT STEP: MONOMER SEQUENCES: E.G. GRADIENT CASE



140°C; [MeOx]<sub>0</sub>: [PhOx]<sub>0</sub>: [MeOTs]<sub>0</sub> = 50:50:1; [MeOx]<sub>0</sub>+ [PhOx]<sub>0</sub> = 3 mol L<sup>-1</sup>; solvent: acetonitrile

# FUTURE OUTLOOK





# CONCLUSIONS

## 1. Modeling & experimental work allow to understand CROP of 2-oxazolines

- chain initiation reactivity: In and low conversion dispersity data
- chain transfer to M reactivity: high monomer conversion dispersity data
- macropropagation reactivity: SEC trace data
- access to absolute MMD data for all macrospecies types
- simple method to assess branching fraction

## 2. The strength is further clear upon the transition to copolymerization with functional M

- FUNC-CLD and derived properties allow for an unbiased qualification
- After parameter tuning all experimental possibilities can be screened
- optimal synthesis conditions can be identified for low and high target DPs

# ACKNOWLEDGEMENTS

1. Special research fund Ghent University
2. FWO Vlaanderen
3. China Scholarship Council

## LABORATORY FOR CHEMICAL TECHNOLOGY

Technologiepark 125, 9052 Ghent, Belgium

E info.lct@ugent.be

T 003293311757

<https://www.lct.ugent.be>